

## **CHEMICAL COMPOUNDS**

### **FIELD OF THE INVENTION**

The present invention relates to benzamide derivatives, compositions and medicaments containing the same, as well as processes for the preparation and use of such compounds, compositions and medicaments. Such benzamide derivatives are useful in the treatment of diseases associated with inappropriate tyrosine and/or serine/threonine kinase activity.

### **BACKGROUND OF THE INVENTION**

An important large family of enzymes is the protein kinase enzyme family. Currently, there are about 500 different known protein kinases. Protein kinases serve to catalyze the phosphorylation of an amino acid side chain in various proteins by the transfer of the  $\gamma$ -phosphate of the ATP-Mg<sup>2+</sup> complex to said amino acid side chain. These enzymes control the majority of the signaling processes inside cells, thereby governing cell function, growth, differentiation and destruction (apoptosis) through reversible phosphorylation of the hydroxyl groups of serine, threonine and tyrosine residues in proteins. Studies have shown that protein kinases are key regulators of many cell functions, including signal transduction, transcriptional regulation, cell motility, and cell division. Several oncogenes have also been shown to encode protein kinases, suggesting that kinases play a role in oncogenesis. These processes are highly regulated, often by complex intermeshed pathways where each kinase will itself be regulated by one or more kinases. Consequently, aberrant or inappropriate protein kinase activity can contribute to the rise of disease states associated with such aberrant kinase activity. Due to their physiological relevance, variety and ubiquitousness, protein kinases have become one of the most important and widely studied family of enzymes in biochemical and medical research.

The protein kinase family of enzymes is typically classified into two main subfamilies: Protein Tyrosine Kinases and Protein Serine/Threonine Kinases, based on the amino acid residue they phosphorylate. The serine/threonine kinases (PSTK), includes cyclic AMP- and cyclic GMP-dependent protein kinases, calcium- and phospholipid-dependent protein kinase, calcium- and calmodulin-dependent protein kinases, casein kinases, cell division cycle protein kinases and others. These kinases are usually cytoplasmic or associated with the particulate fractions of cells, possibly by anchoring proteins. Aberrant protein serine/threonine kinase activity has been implicated or is suspected in a number of pathologies such as rheumatoid arthritis, psoriasis, septic shock, bone loss, many cancers and other proliferative diseases. Accordingly, serine/threonine kinases and the signal transduction pathways which

they are part of are important targets for drug design. The tyrosine kinases phosphorylate tyrosine residues. Tyrosine kinases play an equally important role in cell regulation. These kinases include several receptors for molecules such as growth factors and hormones, including epidermal growth factor receptor, Insulin receptor, platelet derived growth factor receptor and others. Studies have indicated that many tyrosine kinases are transmembrane proteins with their receptor domains located on the outside of the cell and their kinase domains on the inside. Much work is also under progress to identify modulators of tyrosine kinases as well.

A major signal transduction systems utilized by cells is the RhoA- signalling pathways. RhoA is a small GTP binding protein that can be activated by several extracellular stimuli such as growth factor, hormones, mechanic stress, osmotic change as well as high concentration of metabolite like glucose. RhoA activation involves GTP binding, conformation alteration, post-translational modification (geranylgeranyllization and farnesylation) and activation of its intrinsic GTPase activity. Activated RhoA is capable of interacting with several effector proteins including ROCKs and transmit signals into cellular cytoplasm and nucleus.

ROCK1 and 2 constitute a family of kinases that can be activated by RhoA-GTP complex via physical association. Activated ROCKs phosphorylate a number of substrates and play important roles in pivotal cellular functions. The substrates for ROCKs include myosin binding subunit of myosin light chain phosphatase (MBS, also named MYPT1), adducin, moesin, myosin light chain (MLC), LIM kinase as well as transcription factor FHL. The phosphorylation of these substrates modulate the biological activity of the proteins and thus provide a means to alter cell's response to external stimuli. One well documented example is the participation of ROCK in smooth muscle contraction. Upon stimulation by phenylephrine, smooth muscle from blood vessels contracts. Studies have shown that phenylephrine stimulates alpha -adrenergic receptors and leads to the activation of RhoA. Activated RhoA in turn stimulates kinase activity of ROCK1 and which in turn phosphorylates MBS. Such phosphorylation inhibits the enzyme activity of myosin light chain phosphatase and increases the phosphorylation of myosin light chain itself by a calcium-dependent myosin light chain kinase (MLCK) and consequently increases the contractility of myosin-actin bundle, leading to smooth muscle contraction. This phenomena is also sometimes called calcium sensitization. In addition to smooth muscle contraction, ROCKs have also been shown to be involved in cellular functions including apoptosis, cell migration, transcriptional activation, fibrosis, cytokinesis, inflammation and cell proliferation. Moreover, in neurons ROCK plays a critical role in the inhibition of axonal growth by myelin-associated inhibitory factors such as myelin-associated glycoprotein (MAG). ROCK-activity also mediates the collapse of growth cones in developing neurons. Both processes are thought to be mediated by ROCK-induced

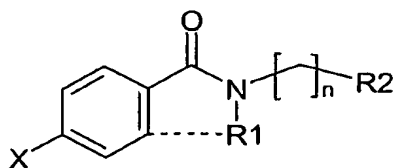
phosphorylation of substrates such as LIM kinase and myosin light chain phosphatase, resulting in increased contractility of the neuronal actin-myosin system.

Inhibitors of ROCKs have been suggested for use in the treatments of a variety of diseases. They include cardiovascular diseases such as hypertension, chronic and congestive heart failure, cardiac hypertrophy, restenosis, chronic renal failure and atherosclerosis. In addition, because of its muscle relaxing properties, it is also suitable for asthma, male erectile dysfunctions, female sexual dysfunction and over-active bladder syndrome. ROCK inhibitors have been shown to possess anti-inflammatory properties. Thus they can be used as treatment for neuroinflammatory diseases such as stroke, multiple sclerosis, Alzheimer's disease, Parkinson's disease, amyotrophic lateral sclerosis and inflammatory pain, as well as other inflammatory diseases such as rheumatoid arthritis, irritable bowel syndrome, inflammatory bowel disease. In addition, based on their neurite outgrowth inducing effects, ROCK inhibitors could be useful drugs for neuronal regeneration, inducing new axonal growth and axonal rewiring across lesions within the CNS. ROCK inhibitors are therefore likely to be useful for regenerative (recovery) treatment of CNS disorders such as spinal cord injury, acute neuronal injury (stroke, traumatic brain injury), Parkinsons disease, Alzheimers disease and other neurodegenerative disorders. Since ROCK inhibitors reduce cell proliferation and cell migration, they could be useful in treating cancer and tumor metastasis. Furthermore, there is evidence suggesting that ROCK inhibitors suppress cytoskeletal rearrangement upon virus invasion, thus they also have potential therapeutic value in anti-viral and anti-bacterial applications. ROCK inhibitors may also be useful for the treatment of insulin resistance and diabetes.

The present inventors have discovered novel benzamide compounds, which are inhibitors of ROCK activity. Such derivatives are therefore potentially useful in the treatment of disorders associated with inappropriate ROCK activity.

**SUMMARY OF THE INVENTION**

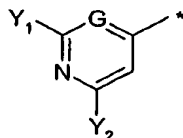
In one aspect of the present invention, there is provided a compound of Formula (I) or a salt, solvate, or physiologically functional derivative thereof:



(I)

wherein:

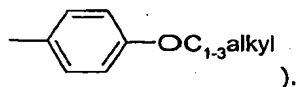
- R1 is hydrogen or C<sub>1-6</sub>alkyl or as indicated by the dotted line is fused to the phenyl forming a 5 or 6 membered ring, optionally containing a double bond;
- n is 0, 1, 2, 3 or 4;
- R2 is aryl, optionally substituted by one or two groups selected from the group consisting of halogen, NH<sub>2</sub>, hydroxy, cyano, C<sub>1-4</sub>alkyl, C<sub>1-4</sub>alkoxy, C<sub>1-4</sub>alkanoyl, haloC<sub>1-4</sub>alkyl, haloC<sub>1-4</sub>alkoxy, aryl, aryloxy, C<sub>1-4</sub>alkoxycarbonyl, C<sub>1-4</sub>alkylsulfonyl and a group R<sub>3</sub>R<sub>4</sub>NSO<sub>2</sub> (wherein R<sub>3</sub> and R<sub>4</sub> are independently hydrogen or C<sub>1-4</sub>alkyl), (CH<sub>2</sub>)<sub>0-3</sub>NHCOOC<sub>1-4</sub>alkyl, and a 5- or 6-membered heteroaryl group;
- or n is 0 and R1 and R2, together with the nitrogen atom to which they are joined, form a 5- or 6-membered monocyclic heterocyclic ring or a 9- or 10-membered bicyclic heterocyclic ring wherein at least the ring which contains the nitrogen atom to which R1 and R2 are joined is non-aromatic, and wherein the 5- or 6-membered monocyclic heterocyclic ring or the 9- or 10-membered bicyclic heterocyclic ring is optionally substituted by one or two groups selected from the group consisting of halogen, hydroxy, cyano, oxo, C<sub>1-4</sub>alkyl, C<sub>1-4</sub>alkanoyl, C<sub>1-4</sub>alkoxy, haloC<sub>1-4</sub>alkyl, haloC<sub>1-4</sub>alkoxy, aryl, aryloxy and C<sub>1-4</sub>alkoxycarbonyl; and
- X is indazolyl, pyrazolyl or a group



wherein

G is CH or N; and

Y<sub>1</sub> and Y<sub>2</sub> are independently hydrogen, halogen and a group NR<sub>5</sub>R<sub>6</sub> (wherein R<sub>5</sub> and R<sub>6</sub> are independently hydrogen, C<sub>1-6</sub>alkyl, C<sub>2-6</sub>alkenyl, or

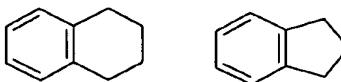


## DETAILED DESCRIPTION OF THE INVENTION

As used herein, the term "C<sub>1-4</sub>alkyl" refers to a straight or branched alkyl which contains one, two, three or four carbon atoms in all isomeric forms. Examples include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl and tert-butyl. As used herein, the term "C<sub>1-6</sub>alkyl" refers to a straight or branched alkyl which contains one, two, three, four, five or six carbon atoms in all isomeric forms. Examples include, in addition to those listed above for C<sub>1-4</sub>alkyl: pentyl, neopentyl, sec-pentyl, n-pentyl, isopentyl, tert-pentyl and hexyl.

As used herein, the term "C<sub>1-4</sub>alkanoyl" refers to an alkanoyl group having from 1 to 4 carbon atoms, such as methanoyl (or "formyl"), ethanoyl (or "acetyl"), propanoyl, isopropanoyl, butanoyl, isobutanoyl and sec-butanoyl.

As used herein, the term "aryl" refers to phenyl or a 8- to 11- membered bicyclic aromatic group wherein one or both rings are aromatic. Examples include phenyl,



indenyl, azulenyl and naphthyl,

As used herein, the term "aryloxy" refers to an aryl group attached via an oxygen atom. Examples of aryloxy include phenyloxy and naphthyloxy,.

As used herein, the term "aryloxyC<sub>1-6</sub>alkyl" refers to an aryloxy group which is attached through a C<sub>1-6</sub>alkylene group. The C<sub>1-6</sub>alkylene group may be in any suitable isomeric form. Examples of aryloxyC<sub>1-6</sub>alkyl include phenoxyethyl.

As used herein, the terms "heteroaryl" and "heteroaromatic group" refer to a 5- or 6-membered monocyclic aromatic group wherein one, two or three carbon atoms are replaced by a heteroatom independently selected from N, O and S, or to a 8- to 11-membered bicyclic aromatic group wherein one to six carbon atoms in total are replaced by a heteroatom independently selected from N, O and S. Examples of 5- or 6-membered heteroaromatic groups include furanyl, thienyl, pyrrolyl, oxazolyl, thiazolyl, imidazolyl, oxadiazolyl, thiadiazolyl, pyridinyl, triazolyl, triazinyl, pyridazyl, pyrimidinyl, isothiazolyl, isoxazolyl, pyrazinyl, pyrazolyl and pyrimidinyl; examples of 8- to 11-membered heteroaromatic groups include indazolyl, quinoxaliny, quinazolinyl, pyridopyrazinyl, benzoxazolyl, benzothiophenyl, benzimidazolyl, naphthyridinyl, quinolinyl, benzofuranyl, indolyl, benzothiazolyl, pyridopyrimidinyl and isoquinolinyl.

As used herein, the terms "heterocyclyl" refers to a 5- or 6-membered non-aromatic cyclic group containing one, two or three heteroatom(s) independently selected from

N, O and S. Examples include pyrrolidinyl, imidazolidinyl, pyrazolidinyl, isothiazolyl, thiazolyl, piperidinyl, piperazinyl, morpholinyl, thiomorpholinyl, tetrahydrofuranyl, dioxolanyl, tetrahydrothienyl, dioxanyl and dithianyl.

As used herein, the term "5- or 6-membered monocyclic heterocyclic ring or a 9- or 10-membered bicyclic heterocyclic ring" refers to a 5- or 6-membered non-aromatic monocyclic heterocyclyl group containing one, two or three heteroatom(s) independently selected from N, O and S, or a 9- or 10-membered bicyclic heterocyclyl group, which contains in total one, two or three heteroatom(s) independently selected from N, O and S, and in which at least one of the rings is non-aromatic. The bicyclic heterocyclic ring may be a fused ring system or a spiro ring system. It should be understood that the 5- or 6-membered monocyclic heterocyclic ring or a 9- or 10-membered bicyclic heterocyclic ring formed by R1 and R2 would be N-linked. Examples of 5- or 6-membered monocyclic heterocyclic rings include pyrrolidinyl, imidazolidinyl, pyrazolidinyl, isothiazolyl, thiazolyl, piperidinyl, piperazinyl, morpholinyl and thiomorpholinyl. Examples of 9- or 10-membered bicyclic heterocyclic rings having a fused structure include tetrahydroisoquinolyl. Examples of 9- or 10-membered bicyclic heterocyclic rings having a spiro structure include triazaspiro[4.5]decanonyl.

As used herein, the term "halogen" refers to fluorine (F), chlorine (Cl), bromine (Br), or iodine (I) and the term "halo" refers to the halogen radicals: fluoro (-F), chloro (-Cl), bromo(-Br), and iodo(-I).

As used herein, the term "C<sub>1-6</sub>alkoxy" refers to a straight chain or branched chain alkoxy (or "alkyloxy") group having from one to six carbon atoms, such as methoxy, ethoxy, propoxy, isopropoxy, butoxy, isobutoxy, sec-butoxy, tert-butoxy, pentoxy, neopentoxy, sec-pentoxy, n-pentoxy, isopentoxy, tert-pentoxy and hexoxy.

As used herein, the term "haloC<sub>1-4</sub>alkyl" refers to a halogen-substituted C<sub>1-4</sub>alkyl group such as -CF<sub>3</sub>. Similarly, the term "haloC<sub>1-4</sub>alkoxy" refers to a halogen-substituted C<sub>1-4</sub>alkoxy group such as CF<sub>3</sub>O-.

As used herein, the term "C<sub>1-4</sub>alkoxycarbonyl" refers to the group (C<sub>1-4</sub>alkyl)OC(=O)-. Examples of C<sub>1-4</sub>alkoxycarbonyl include ethyloxycarbonyl (C<sub>2</sub>H<sub>5</sub>OC(=O)-) and methyloxycarbonyl (CH<sub>3</sub>OC(=O)-).

As used herein, the term "C<sub>2-6</sub>alkenyl" refers to a hydrocarbon radical having from two to six carbons and at least one carbon-carbon double bond. Examples of "C<sub>2-6</sub>alkenyl" include ethenyl, propenyl, butenyl, 2-butenyl, and isobutenyl.

As used herein, the term "salt" refers to any salt of a compound according to the present invention prepared from an inorganic or organic acid or base, quaternary ammonium salts and internally formed salts. Physiologically acceptable salts are particularly suitable for medical applications because of their greater aqueous solubility relative to the parent compounds. Such salts must clearly have a physiologically acceptable anion or cation. Suitably physiologically acceptable salts of the compounds of the present invention include acid addition salts formed with inorganic acids such as hydrochloric, hydrobromic, hydroiodic, phosphoric, metaphosphoric, nitric and sulfuric acids, and with organic acids, such as tartaric, acetic, trifluoroacetic, citric, malic, lactic, fumaric, benzoic, formic, propionic, glycolic, gluconic, maleic, succinic, camphorsulfuric, isothionic, mucic, gentisic, isonicotinic, saccharic, glucuronic, furoic, glutamic, ascorbic, anthranilic, salicylic, phenylacetic, mandelic, embonic (pamoic), methanesulfonic, ethanesulfonic, pantothenic, stearic, sulfinitic, alginic, galacturonic and arylsulfonic, for example benzenesulfonic and p-toluenesulfonic, acids; base addition salts formed with alkali metals and alkaline earth metals and organic bases such as N,N-dibenzylethylenediamine, chloroprocaine, choline, diethanolamine, ethylenediamine, meglumine (N-methylglucamine), lysine and procaine; and internally formed salts. Salts having a non-physiologically acceptable anion or cation are within the scope of the invention as useful intermediates for the preparation of physiologically acceptable salts and/or for use in non-therapeutic, for example, *in vitro*, situations.

As used herein, the term "solvate" refers to a complex of variable stoichiometry formed by a solute (in this invention, a compound of formula (I) or formula (Ia), or a salt or physiologically functional derivative thereof) and a solvent. Such solvents for the purpose of the invention may not interfere with the biological activity of the solute. Examples of suitable solvents include, but are not limited to, water, methanol, ethanol and acetic acid. Preferably the solvent used is a pharmaceutically acceptable solvent. Examples of suitable pharmaceutically acceptable solvents include water, ethanol and acetic acid. Most preferably the solvent used is water.

As used herein, the term "physiologically functional derivative" refers to any pharmaceutically acceptable derivative of a compound of the present invention, for example, an ester or an amide, which upon administration to a mammal is capable of providing (directly or indirectly) a compound of the present invention or an active metabolite thereof. Such derivatives are clear to those skilled in the art, without undue experimentation, and with reference to the teaching of Burger's Medicinal Chemistry And Drug Discovery, 5<sup>th</sup> Edition, Vol 1: Principles and Practice, which is incorporated herein by reference to the extent that it teaches physiologically functional derivatives.

As used herein, the term "substituted" refers to substitution with the named substituent or substituents, multiple degrees of substitution being allowed unless otherwise stated.

In one embodiment, R1 is hydrogen.

In one embodiment, n is 1 or 2.

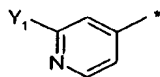
In one embodiment, R2 is aryl (such as phenyl or naphthyl), optionally substituted by one or two groups selected from the group consisting of halogen and C<sub>1-4</sub>alkoxy (such as methoxy or ethoxy).

In another embodiment, n is 0 and R1 and R2, together with the nitrogen atom to which they are joined, form a 6-membered monocyclic heterocyclic ring (such as piperidiny1 or piperaziny1) or a 10-membered bicyclic heterocyclic ring wherein at least the ring which each contains the nitrogen atom to which R1 and R2 are joined is non-aromatic (such as tetrahydroisoquinoliny1 or triazaspiro[4.5]decanony1), wherein the 6-membered monocyclic heterocyclic ring or 10-membered bicyclic heterocyclic ring are both optionally substituted by one or two groups selected from oxo, C<sub>1-4</sub>alkyl (such as methyl or ethyl), phenyl and C<sub>1-4</sub>alkoxycarbonyl (such as ethyloxycarbonyl or methyloxycarbonyl).

In one embodiment, X is indazolyl, such as 1-H-indazol-5-yl.

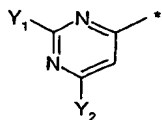
In another embodiment, X is pyrazolyl, such as 1H-pyrazol-4-yl.

In another embodiment, X is a group:



wherein Y<sub>1</sub> is hydrogen or halogen (such as chloro).

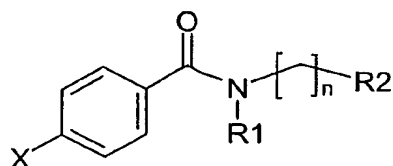
In another embodiment, X is a group:





wherein one of  $Y_1$  and  $Y_2$  is hydrogen, and the other is hydrogen, halogen or a group  $NR_5R_6$  wherein  $R_5$  and  $R_6$  are independently hydrogen,  $C_{1-6}$ alkyl (such as methyl or ethyl) or  $C_{2-6}$ alkenyl (such as allyl).

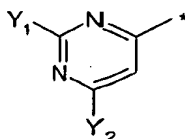
In another aspect, the present invention provides a compound of Formula (Ia) or a salt, solvate, or physiologically functional derivative thereof:



(Ia)

wherein

- $R_1$  is hydrogen;
- $n$  is 1 or 2;
- $R_2$  is aryl, optionally substituted by one or two groups selected from the group consisting of halogen and  $C_{1-4}$ alkoxy;
- or  $n$  is 0 and  $R_1$  and  $R_2$ , together with the nitrogen atom to which they are joined, form a 6-membered monocyclic heterocyclic ring or a 10-membered bicyclic heterocyclic ring wherein the 6-membered monocyclic heterocyclic ring or the 10-membered bicyclic heterocyclic ring are optionally substituted by one or two groups selected from oxo,  $C_{1-4}$ alkyl, phenyl and  $C_{1-4}$ alkoxycarbonyl;
- $X$  is indazolyl, pyrazolyl, 4-pyridinyl or a group



wherein  $Y_1$  and  $Y_2$  are independently hydrogen, halogen and a group  $NR_5R_6$  (wherein  $R_5$  and  $R_6$  are independently hydrogen,  $C_{1-6}$ alkyl or  $C_{2-6}$ alkenyl).

In one embodiment of formula (Ia),  $R_2$  is phenyl, optionally substituted by one or two  $C_{1-4}$ alkoxy (such as methoxy or ethoxy).

In another embodiment of formula (Ia),  $n$  is 0 and  $R_1$  and  $R_2$ , together with the nitrogen atom to which they are joined, form piperidinyl, piperazinyl, tetrahydroisoquinolinyl or triazaspiro[4.5]decanonyl, wherein the 6-membered

monocyclic heterocyclic ring or 10-membered bicyclic heterocyclic ring are both optionally substituted by one or two groups selected from oxo, C<sub>1-4</sub>alkyl (such as methyl or ethyl), phenyl and C<sub>1-4</sub>alkoxycarbonyl (such as ethyloxycarbonyl or methyloxycarbonyl).

In one embodiment of formula (Ia), X is 1-H-indazol-5-yl, 1H-pyrazol-4-yl, 4-pyridinyl, 2-amino-4-pyrimidinyl, 6-allylamino-4-pyrimidinyl or 6-amino-4-pyrimidinyl.

Specific examples of compounds of the present invention include:

N-benzyl-4-(4-pyridinyl)benzamide  
N-(2-phenylethyl)-4-(4-pyridinyl)benzamide  
N-(3-methoxybenzyl)-4-(4-pyridinyl)benzamide  
N-(3-methoxybenzyl)-4-(1H-pyrazol-4-yl)benzamide  
4-(2-chloro-4-pyridinyl)-N-(3-methoxybenzyl)benzamide  
4-(2-amino-4-pyrimidinyl)-N-(3-methoxybenzyl)benzamide  
N-(3-methoxybenzyl)-4-(4-pyrimidinyl)benzamide  
4-[6-(allylamino)-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide  
4-[6-amino-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide  
4-(1H-indazol-5-yl)-N-(3-methoxybenzyl)benzamide  
4-(2-amino-4-pyrimidinyl)-N-[[3-(methyloxy)phenyl]methyl]benzamide  
4-(2-[[4-(methyloxy)phenyl]amino]-4-pyrimidinyl)-N-[[3-(methyloxy)phenyl]methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(2-chlorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(4-fluorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(4-chlorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(2-fluorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[[2-(methyloxy)phenyl]methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(2-methylphenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[[4-(methyloxy)phenyl]methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-(2,3-dihydro-1H-inden-1-yl)benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(1R)-1,2,3,4-tetrahydro-1-naphthalenyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[[3-(trifluoromethyl)phenyl]methyl]benzamide  
1,1-dimethylethyl({3-[[4-(2-amino-4-pyrimidinyl)phenyl]carbonyl]amino)methyl}phenyl)methyl carbamate  
4-(2-amino-4-pyrimidinyl)-N-[(3-bromophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(3-chlorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(3-fluorophenyl)methyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-(phenylmethyl)benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(1S)-1-[4-(methyloxy)phenyl]ethyl]benzamide  
4-(2-amino-4-pyrimidinyl)-N-[(1S)-1-phenylpropyl]benzamide  
6-(2-amino-4-pyrimidinyl)-2-(phenylmethyl)-1(2H)-isoquinolinone  
4-(2-amino-4-pyrimidinyl)-N-[(3-hydroxyphenyl)methyl]benzamide

4-(2-amino-4-pyrimidinyl)-N-({3-[(difluoromethoxy)phenyl]methyl}benzamide  
 4-(2-amino-4-pyrimidinyl)-N-(1-methyl-1-phenylethyl)benzamide  
 4-(2-amino-4-pyrimidinyl)-N-[(3,5-dichlorophenyl)methyl]benzamide  
 4-(2-amino-4-pyrimidinyl)-N-(4-biphenylmethyl)benzamide  
 4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-phenylethyl]benzamide  
 1,1-dimethylethyl{3-[1-({4-(2-amino-4-pyrimidinyl)phenyl}carbonyl)amino)ethyl]phenyl}carbamate  
 N-[(2-aminophenyl)methyl]-4-(2-amino-4-pyrimidinyl)benzamide  
 4-(2-amino-4-pyrimidinyl)-N-[(1S)-1,2,3,4-tetrahydro-1-naphthalenyl]benzamide  
 5-(2-amino-4-pyrimidinyl)-2-(phenylmethyl)-2,3-dihydro-1H-isoindol-1-one  
 4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-[3-(methyloxy)phenyl]ethyl]benzamide  
 4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-phenylpropyl]benzamide

and their salts, solvates and physiologically functional derivatives thereof.

The compounds of formulae (I) and (Ia) have the ability to crystallise in more than one form, a characteristic, which is known as polymorphism, and it is understood that such polymorphic forms ("polymorphs") are within the scope of formulae (I) and (Ia). Polymorphism generally can occur as a response to changes in temperature or pressure or both and can also result from variations in the crystallisation process. Polymorphs can be distinguished by various physical characteristics known in the art such as x-ray diffraction patterns, solubility, and melting point.

Certain of the compounds described herein may exist in stereoisomeric forms (i.e. they may contain one or more asymmetric carbon atoms or may exhibit *cis-trans* isomerism). The individual stereoisomers (enantiomers and diastereoisomers) and mixtures of these are included within the scope of the present invention. Likewise, it is understood that compounds of formulae (I) and (Ia) may exist in tautomeric forms other than that shown in the formulae and these are also included within the scope of the present invention.

As referred to above, individual enantiomers of compounds of formulae (I) and (Ia) may be prepared and an indication of the preferred stereochemistry for such enantiomers has been given. In a preferred embodiment, an optically pure enantiomer is desired. The term "optically pure enantiomer" means that the compound contains greater than about 90 % of the desired isomer by weight, preferably greater than about 95 % of the desired isomer by weight, and most preferably greater than about 99 % of the desired isomer by weight, said weight percent based upon the total weight of the isomer(s) of the compound.

It is to be understood that the following embodiments refer to compounds within the scope of both formula (I) and formula (Ia) as defined above unless specifically limited

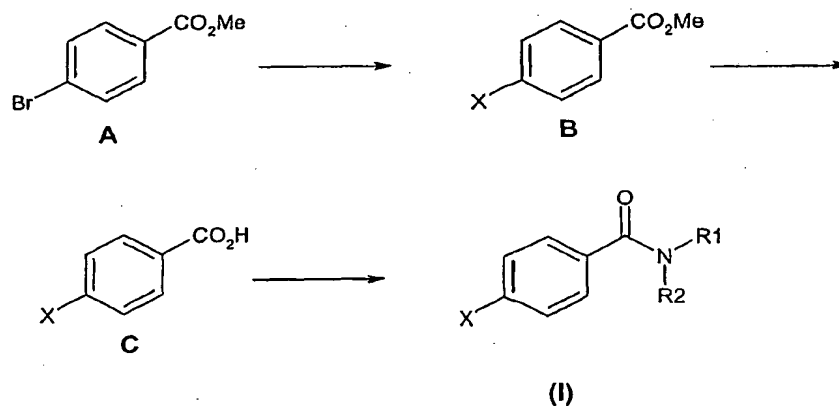
by the definition of each formula or specifically limited otherwise. It is also understood that the embodiments of the present invention described herein, including uses and compositions, are applicable to both formula (I) and formula (Ia).

The compounds of this invention may be made by a variety of methods, including standard chemistry. Any previously defined variable will continue to have the previously defined meaning unless otherwise indicated. Illustrative general synthetic methods are set out below and then specific compounds of the invention are prepared in the Working Examples.

Compounds of general formula (I) may be prepared by methods known in the art of organic synthesis as set forth in part by the following synthesis schemes. In all of the schemes described below, it is well understood that protecting groups for sensitive or reactive groups are employed where necessary in accordance with general principles of chemistry. Protecting groups are manipulated according to standard methods of organic synthesis (T. W. Green and P. G. M. Wuts (1991) Protecting Groups in Organic Synthesis, John Wiley & Sons). These groups are removed at a convenient stage of the compound synthesis using methods that are readily apparent to those skilled in the art. The selection of processes as well as the reaction conditions and order of their execution shall be consistent with the preparation of compounds of Formula (I).

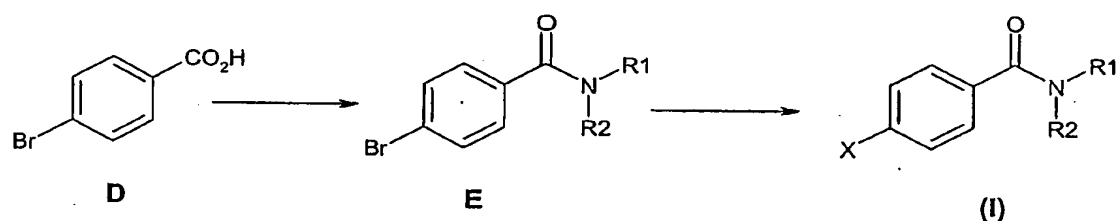
The compounds of formula (I) and (Ia) may be made by a variety of methods, including standard chemistry. Any previously defined variable will continue to have the previously defined meaning unless otherwise indicated. Illustrative general synthetic methods are in Schemes 1 and 2 and 3.

Scheme 1



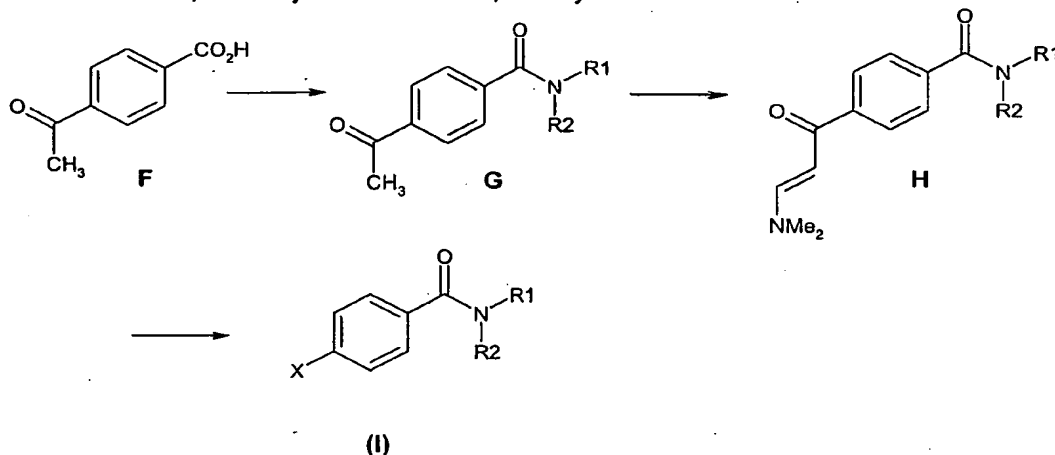
As illustrated in Scheme 1, compounds of general formula (I) may be synthesized starting with compound A, methyl 4-bromobenzoate. Compound A can be coupled in a Suzuki reaction with an appropriate heteroarylboronic acid or heteroarylboronate ester at temperatures between 25 and 250 °C, often in the presence of an appropriate additive, to give B. For example, reaction of A with 4-pyridylboronic acid in 1,2-dimethoxyethane : ethanol (2:1) with dichlorobis-(triphenylphosphine)palladium(II) and 2M aqueous sodium carbonate at 175°C for 10 minutes under microwave irradiation provides compound B. Intermediate B can be hydrolyzed to the acid using any of the standard ester hydrolysis procedures known to those skilled in the art. For example, treatment of Compound B with 1N aqueous lithium hydroxide : 1,4-dioxane (1:1) at ambient temperature for 2 days gives carboxylic acid C. Coupling of the acid C with an amine using standard amide bond forming reactions known to those skilled in the art, provides Compounds of formula (I). For example, activation of Compound C with polystyrene-supported carbodiimide and HOBT in DMF, followed by addition of an amine and stirring for several hours at ambient temperature provides Compounds of formula (I).

Scheme 2



As shown in Scheme 2, compounds of general formula (I) can be synthesized from compounds of general formula D by first coupling of the carboxylic acid D with an amine via a number of useful amide coupling reactions known to those skilled in the art to give compounds of general formula E. For example, compound D when activated with EDC hydrochloride and HOBT in DMF can be coupled with an amine to give compounds of formula E. Compounds of formula E can be converted to compounds of formula (I) through a variety of metal mediated coupling reactions well known to those skilled in the art. For example, reaction of aryl halides such as E with an aryl tin species or an aryl boronic acid species can be carried out in an appropriate solvent in the presence of an appropriate catalyst and an appropriate base at a temperature between 30 °C and 250 °C. These reactions (Suzuki reaction with an aryl boronic acid and Stille reaction with an aryl tin reagent) are well described in the literature, and a number of catalyst, base, solvent, and temperature combinations have proven useful. For example, heating an appropriate compound of general formula E with an aryl boronic acid, aqueous sodium carbonate and

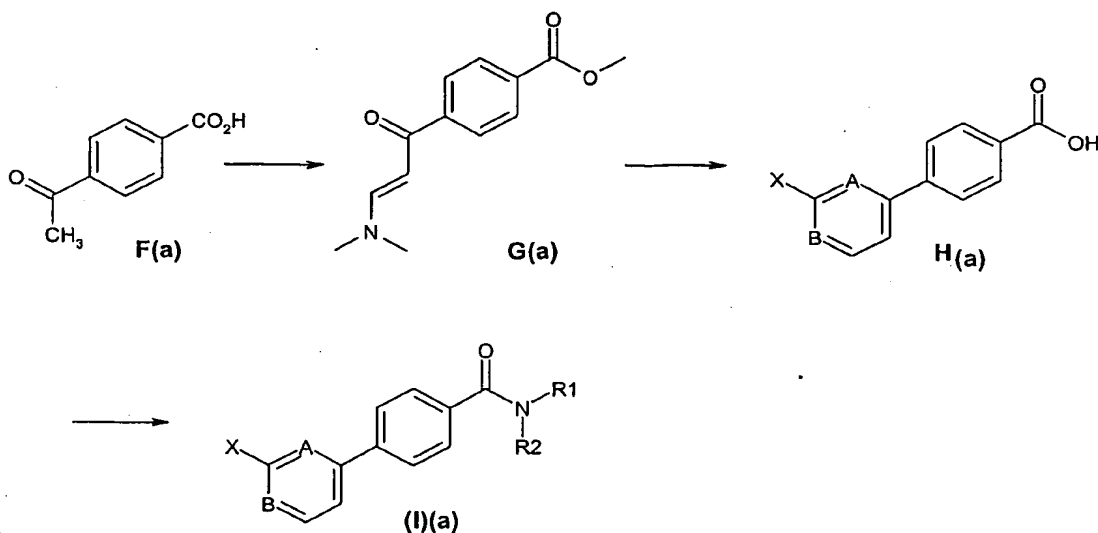
dichlorobis(triphenylphosphine) palladium (II) in dimethoxyethane at 150 °C for 10 minutes in a SmithSynthesizer microwave is one method useful for synthesis of products of general formula (I). Other well described reactions such as the Heck reaction, Sonogashira reaction, carbonylation reactions and cyanation reactions may be used to generate other compounds of general formula (I) that replace the bromine of compounds E with different functionality, such as substituted olefins, substituted acetylenes, substituted amides, a carboxylic acid, or nitrile. For all of these types of reactions, a number of catalyst, base, solvent, and temperature combinations have been explored and have proven useful for carrying out the desired transformation. Like the Suzuki and Stille reactions, a number of catalyst, base, solvent, and temperature combinations have proven useful to carry out the Sonogashira reactions, Heck reactions, carbonylation reactions, and cyanations.



### Scheme 3

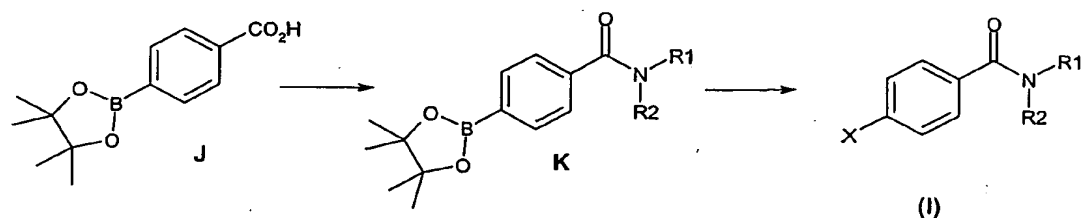
Scheme 3 depicts an alternate way to synthesize compounds of general formula (I). Compound **G** can be synthesized by reaction of compound **F** using a suitable amide coupling reaction in a suitable solvent at a suitable temperature. There are a variety of conditions known in the chemical literature that are useful for amide coupling reactions of a benzoic acid with an amine as outlined in Scheme 3 above. Compounds of formula **H** can be prepared from compounds of formula **G** by heating with dimethylformamide dimethylacetal. Application of these sorts of conditions, as described above and further illustrated in the detailed examples following, give compounds of general formula **H**. Compounds of general formula **H** can be converted into compounds of general formula (I) by condensation of the enaminoketone with a suitable bis-nucleophile and cyclization to a heterocycle. For example, reaction of **H** with a bis-nucleophile in the presence of a strong base in an appropriate solvent at temperatures between 30 and 250 °C will give compounds of formula (I). In particular, reaction of **H** with guanidine hydrochloride in the presence of sodium ethoxide in refluxing ethanol will give compounds of formula (I).

Scheme 3a is an embodiment of Scheme 3



#### Scheme 3a

Scheme 3 depicts an alternate way to synthesize compounds of general formula (I). Compounds of formula **G(a)** can be prepared from compounds of formula **F(a)** by heating with dimethylformamide dimethylacetal. Compounds of general formula **G(a)** can be converted into compounds of general formula **H(a)** by condensation of the enaminoketone with a suitable bis-nucleophile and cyclization to a heterocycle. For example, reaction of **G(a)** with a bis-nucleophile in the presence of a strong base in an appropriate solvent at temperatures between 30 and 250 °C will give compounds of formula **H(a)**. In particular, reaction of **G(a)** with guanidine hydrochloride in the presence of potassium carbonate in refluxing 2-methoxyethanol will give compounds of formula **H(a)**. Compounds of general formula **(I)(a)** can be synthesized by reaction of compounds of general formula **H(a)** using a suitable amide coupling reaction in a suitable solvent at a suitable temperature. There are a variety of conditions known in the chemical literature that are useful for amide coupling reactions of a benzoic acid with an amine as outlined in Scheme 3a above.

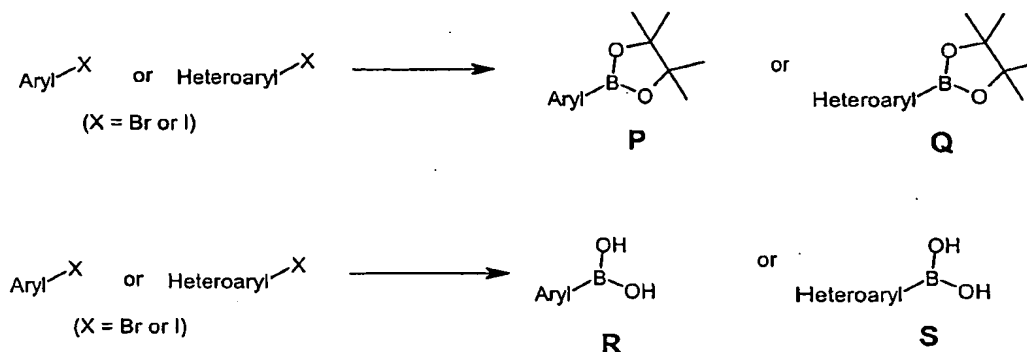


#### Scheme 4

Scheme 4 illustrates the use of a phenylboronate ester in a Suzuki coupling reaction to give compounds of formula (I). Compound K can be prepared from Compound J, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzoic acid, by a suitable amide coupling reaction in a suitable solvent at a suitable temperature. Compounds of formula K can be prepared from compounds of formula J by the previously described amide coupling procedure (see Scheme 2 above). Compounds of formula (I) can be prepared from Compounds of formula K using the previously described Suzuki reaction procedure (see Scheme 2 above).

In some instances, the chemistry depicted in schemes 1 and 2 and 4 describe the synthesis of compounds of general formula (I) which make use of boronic acids or boronate esters. Many boronic acids and boronate esters are commercially available. When not commercially available, boronic acids and boronate esters may be synthesized by standard methods, including those depicted in scheme 5 (see below). Aryl or heteroaryl boronate esters may be synthesized by reaction of an aryl or heteroaryl halide with bis(pinacolato)diboron and an appropriate palladium catalyst in an appropriate solvent with appropriate additives. For example, reaction of an aryl halide and bis(pinacolato)diboron with  $\text{PdCl}_2(\text{dppf})_2$ , and potassium acetate, in DMF as solvent at 80 °C for 90 minutes can give boronate esters of general formula P. Aryl or heteroaryl boronic acids may be synthesized by treating an appropriate aryl halide or heteroaryl halide with a strong base such as n-BuLi or t-BuLi in a solvent such as THF or dioxane, followed by reaction of the intermediate organometallic species with a reagent to introduce the boron. For example, reaction of an aryl halide in THF at -70 °C with n-butyl lithium, followed by addition of tri-isopropylborate gives, after standard work up, aryl boronic acids of general formula R.





### Scheme 5

Intermediates used in schemes 1, 2, 3 and 4 can be obtained from commercial sources or synthesized by one skilled in the art. Some of the intermediates may be synthesized, for example, by the synthetic sequences outlined in scheme 5 and further detailed in the experimental sections following.

As mentioned above, the compounds of the present invention are inhibitors of ROCK activity which are useful in the treatment of disorders associated with inappropriate ROCK activity. Thus, in a further aspect of the present invention, there is provided a compound of formula (I), or a salt, solvate, or a physiologically functional derivative thereof for use in therapy.

In a further aspect of the present invention, there is provided a method of treating a disorder in a mammal, said disorder being mediated by inappropriate ROCK-1 activity, comprising: administering to said mammal a therapeutically effective amount of a compound of formula (I) or a salt, solvate or a physiologically functional derivative thereof.

In a further aspect of the present invention, there is provided the use of a compound of formula (I), or a salt, solvate, or a physiologically functional derivative thereof in the preparation of a medicament for use in the treatment of a disorder mediated by inappropriate ROCK-1 activity.

The term "disorder mediated by inappropriate ROCK-1 activity" includes cardiovascular diseases (such as hypertension, chronic and congestive heart failure, cardiac hypertrophy, restenosis, chronic renal failure and atherosclerosis); asthma, male erectile dysfunctions, female sexual dysfunction and over-active bladder syndrome; neuroinflammatory diseases (such as stroke, multiple sclerosis,

Alzheimer's disease, Parkinson's disease, amyotrophic lateral sclerosis and inflammatory pain); other inflammatory diseases (such as rheumatoid arthritis, irritable bowel syndrome and inflammatory bowel disease); spinal cord injury, acute neuronal injury (stroke, traumatic brain injury), Parkinson's disease, Alzheimer's disease and other neurodegenerative disorders; cancer and tumor metastasis; viral and bacterial diseases; and diabetes.

While it is possible that, for use in therapy, therapeutically effective amounts of a compound of formula (I), as well as salts, solvates and physiological functional derivatives thereof, may be administered as the raw chemical, it is possible to present the active ingredient as a pharmaceutical composition. Accordingly, the invention further provides a pharmaceutical composition, comprising a therapeutically effective amount of a compound of formula (I), or a physiologically functional derivative thereof and one or more of pharmaceutically acceptable carriers, diluents and excipients. The compounds of the formula (I) and salts, solvates and physiological functional derivatives thereof, are as described above. The carrier(s), diluent(s) or excipient(s) must be acceptable in the sense of being compatible with the other ingredients of the formulation and not deleterious to the recipient thereof. In accordance with another aspect of the invention there is also provided a process for the preparation of a pharmaceutical formulation including admixing a compound of the formula (I), or salts, solvates and physiological functional derivatives thereof, with one or more pharmaceutically acceptable carriers, diluents or excipients.

Pharmaceutical formulations may be presented in unit dose forms containing a predetermined amount of active ingredient per unit dose. Such a unit may contain, for example, 0.5mg to 1g, preferably 1mg to 700mg, more preferably 5mg to 100mg of a compound of the formula (I), depending on the condition being treated, the route of administration and the age, weight and condition of the patient, or pharmaceutical formulations may be presented in unit dose forms containing a predetermined amount of active ingredient per unit dose. Preferred unit dosage formulations are those containing a daily dose or sub-dose, as herein above recited, or an appropriate fraction thereof, of an active ingredient. Furthermore, such pharmaceutical formulations may be prepared by any of the methods well known in the pharmacy art.

Pharmaceutical formulations may be adapted for administration by any appropriate route, for example by the oral (including buccal or sublingual), rectal, nasal, topical (including buccal, sublingual or transdermal), vaginal or parenteral (including subcutaneous, intramuscular, intravenous or intradermal) route. Such formulations may be prepared by any method known in the art of pharmacy, for example by bringing into association the active ingredient with the carrier(s) or excipient(s).

Pharmaceutical formulations adapted for oral administration may be presented as discrete units such as capsules or tablets; powders or granules; solutions or suspensions in aqueous or non-aqueous liquids; edible foams or whips; or oil-in-water liquid emulsions or water-in-oil liquid emulsions.

For instance, for oral administration in the form of a tablet or capsule, the active drug component can be combined with an oral, non-toxic pharmaceutically acceptable inert carrier such as ethanol, glycerol, water and the like. Powders are prepared by comminuting the compound to a suitable fine size and mixing with a similarly comminuted pharmaceutical carrier such as an edible carbohydrate, as, for example, starch or mannitol. Flavoring, preservative, dispersing and coloring agent can also be present.

Capsules are made by preparing a powder mixture, as described above, and filling formed gelatin sheaths. Glidants and lubricants such as colloidal silica, talc, magnesium stearate, calcium stearate or solid polyethylene glycol can be added to the powder mixture before the filling operation. A disintegrating or solubilizing agent such as agar-agar, calcium carbonate or sodium carbonate can also be added to improve the availability of the medicament when the capsule is ingested.

Moreover, when desired or necessary, suitable binders, lubricants, disintegrating agents and coloring agents can also be incorporated into the mixture. Suitable binders include starch, gelatin, natural sugars such as glucose or beta-lactose, corn sweeteners, natural and synthetic gums such as acacia, tragacanth or sodium alginate, carboxymethylcellulose, polyethylene glycol, waxes and the like. Lubricants used in these dosage forms include sodium oleate, sodium stearate, magnesium stearate, sodium benzoate, sodium acetate, sodium chloride and the like. Disintegrators include, without limitation, starch, methyl cellulose, agar, bentonite, xanthan gum and the like. Tablets are formulated, for example, by preparing a powder mixture, granulating or slugging, adding a lubricant and disintegrant and pressing into tablets. A powder mixture is prepared by mixing the compound, suitably comminuted, with a diluent or base as described above, and optionally, with a binder such as carboxymethylcellulose, an algininate, gelatin, or polyvinyl pyrrolidone, a solution retardant such as paraffin, a resorption accelerator such as a quaternary salt and/or an absorption agent such as bentonite, kaolin or dicalcium phosphate. The powder mixture can be granulated by wetting with a binder such as syrup, starch paste, acacia mucilage or solutions of cellulosic or polymeric materials and forcing through a screen. As an alternative to granulating, the powder mixture can be run through the tablet machine and the result is imperfectly formed slugs broken into granules. The granules can be lubricated to prevent sticking to the tablet forming dies by means of the addition of stearic acid, a stearate salt, talc or mineral oil. The lubricated mixture is then compressed into tablets. The compounds of the

present invention can also be combined with a free flowing inert carrier and compressed into tablets directly without going through the granulating or slugging steps. A clear or opaque protective coating consisting of a sealing coat of shellac, a coating of sugar or polymeric material and a polish coating of wax can be provided. Dyestuffs can be added to these coatings to distinguish different unit dosages.

Oral fluids such as solution, syrups and elixirs can be prepared in dosage unit form so that a given quantity contains a predetermined amount of the compound. Syrups can be prepared by dissolving the compound in a suitably flavored aqueous solution, while elixirs are prepared through the use of a non-toxic alcoholic vehicle. Suspensions can be formulated by dispersing the compound in a non-toxic vehicle. Solubilizers and emulsifiers such as ethoxylated isostearyl alcohols and polyoxy ethylene sorbitol ethers, preservatives, flavor additive such as peppermint oil or natural sweeteners or saccharin or other artificial sweeteners, and the like can also be added.

Where appropriate, dosage unit formulations for oral administration can be microencapsulated. The formulation can also be prepared to prolong or sustain the release as for example by coating or embedding particulate material in polymers, wax or the like.

The compounds of formula (I), and salts, solvates and physiological functional derivatives thereof, can also be administered in the form of liposome delivery systems, such as small unilamellar vesicles, large unilamellar vesicles and multilamellar vesicles. Liposomes can be formed from a variety of phospholipids, such as cholesterol, stearylamine or phosphatidylcholines.

The compounds of formula (I) and salts, solvates and physiological functional derivatives thereof may also be delivered by the use of monoclonal antibodies as individual carriers to which the compound molecules are coupled. The compounds may also be coupled with soluble polymers as targetable drug carriers. Such polymers can include polyvinylpyrrolidone, pyran copolymer, polyhydroxypropylmethacrylamide -phenol, polyhydroxyethylaspartamidophenol, or polyethyleneoxidepolylysine substituted with palmitoyl residues. Furthermore, the compounds may be coupled to a class of biodegradable polymers useful in achieving controlled release of a drug, for example, polylactic acid, polyepsilon caprolactone, polyhydroxy butyric acid, polyorthoesters, polyacetals, polydihydropyrans, polycyanoacrylates and cross-linked or amphipathic block copolymers of hydrogels.

Pharmaceutical formulations adapted for transdermal administration may be presented as discrete patches intended to remain in intimate contact with the epidermis of the recipient for a prolonged period of time. For example, the active

ingredient may be delivered from the patch by iontophoresis as generally described in *Pharmaceutical Research*, 3(6), 318 (1986).

Pharmaceutical formulations adapted for topical administration may be formulated as ointments, creams, suspensions, lotions, powders, solutions, pastes, gels, sprays, aerosols or oils.

For treatments of the eye or other external tissues, for example mouth and skin, the formulations are preferably applied as a topical ointment or cream. When formulated in an ointment, the active ingredient may be employed with either a paraffinic or a water-miscible ointment base. Alternatively, the active ingredient may be formulated in a cream with an oil-in-water cream base or a water-in-oil base.

Pharmaceutical formulations adapted for topical administrations to the eye include eye drops wherein the active ingredient is dissolved or suspended in a suitable carrier, especially an aqueous solvent.

Pharmaceutical formulations adapted for topical administration in the mouth include lozenges, pastilles and mouth washes.

Pharmaceutical formulations adapted for rectal administration may be presented as suppositories or as enemas.

Pharmaceutical formulations adapted for nasal administration wherein the carrier is a solid include a coarse powder having a particle size for example in the range 20 to 500 microns which is administered in the manner in which snuff is taken, i.e. by rapid inhalation through the nasal passage from a container of the powder held close up to the nose. Suitable formulations wherein the carrier is a liquid, for administration as a nasal spray or as nasal drops, include aqueous or oil solutions of the active ingredient.

Pharmaceutical formulations adapted for administration by inhalation include fine particle dusts or mists, which may be generated by means of various types of metered, dose pressurised aerosols, nebulizers or insufflators.

Pharmaceutical formulations adapted for vaginal administration may be presented as pessaries, tampons, creams, gels, pastes, foams or spray formulations.

Pharmaceutical formulations adapted for parenteral administration include aqueous and non-aqueous sterile injection solutions which may contain anti-oxidants, buffers, bacteriostats and solutes which render the formulation isotonic with the blood of the intended recipient; and aqueous and non-aqueous sterile suspensions which may

include suspending agents and thickening agents. The formulations may be presented in unit-dose or multi-dose containers, for example sealed ampoules and vials, and may be stored in a freeze-dried (lyophilized) condition requiring only the addition of the sterile liquid carrier, for example water for injections, immediately prior to use. Extemporaneous injection solutions and suspensions may be prepared from sterile powders, granules and tablets.

It should be understood that in addition to the ingredients particularly mentioned above, the formulations may include other agents conventional in the art having regard to the type of formulation in question, for example those suitable for oral administration may include flavouring agents.

A therapeutically effective amount of a compound of the present invention will depend upon a number of factors including, for example, the age and weight of the human or other animal, the precise condition requiring treatment and its severity, the nature of the formulation, and the route of administration, and will ultimately be at the discretion of the attendant physician or veterinarian. However, an effective amount of a compound of formula (I) for the treatment of neoplastic growth, for example colon or breast carcinoma, will generally be in the range of 0.1 to 100 mg/kg body weight of recipient (mammal) per day and more usually in the range of 1 to 10 mg/kg body weight per day. Thus, for a 70kg adult mammal, the actual amount per day would usually be from 70 to 700 mg and this amount may be given in a single dose per day or more usually in a number (such as two, three, four, five or six) of sub-doses per day such that the total daily dose is the same. An effective amount of a salt or solvate, or physiologically functional derivative thereof, may be determined as a proportion of the effective amount of the compound of formula (I) per se. It is envisaged that similar dosages would be appropriate for treatment of the other conditions referred to above.

## EXAMPLES

As used herein the symbols and conventions used in these processes, schemes and examples are consistent with those used in the contemporary scientific literature, for example, the *Journal of the American Chemical Society* or the *Journal of Biological Chemistry*. Standard single-letter or three-letter abbreviations are generally used to designate amino acid residues, which are assumed to be in the L-configuration unless otherwise noted. Unless otherwise noted, all starting materials were obtained from commercial suppliers and used without further purification. Specifically, the following abbreviations may be used in the examples and throughout the specification:

g (grams);	mg (milligrams);
L (liters);	mL (milliliters);
μL (microliters);	psi (pounds per square inch);
M (molar);	mM (millimolar);
i. v. (intravenous);	Hz (Hertz);
MHz (megaHertz);	mol (moles);
mmol (millimoles);	rt (room temperature);
min (minutes);	h (hours);
mp (melting point);	TLC (thin layer chromatography);
T <sub>r</sub> (retention time);	RP (reverse phase);
MeOH (methanol);	<i>i</i> -PrOH (isopropanol);
TEA (triethylamine);	TFA (trifluoroacetic acid);
TFAA (trifluoroacetic anhydride);	THF (tetrahydrofuran);
DMSO (dimethylsulfoxide);	AcOEt (ethyl acetate);
DME (1,2-dimethoxyethane);	DCM (dichloromethane);
DCE (dichloroethane);	DMF ( <i>N,N</i> -dimethylformamide);
DMPU ( <i>N,N'</i> -dimethylpropyleneurea);	CDI (1,1'-carbonyldiimidazole);
IBCF (isobutyl chloroformate);	HOAc (acetic acid);
HOSu ( <i>N</i> -hydroxysuccinimide);	HOBT (1-hydroxybenzotriazole);
mCPBA (meta-chloroperbenzoic acid);	
EDC (1-[(3-dimethylamino) propyl]-3-ethylcarbodiimide hydrochloride);	
BOC ( <i>tert</i> -butoxycarbonyl);	Fmoc (9-fluorenylmethoxycarbonyl);
DCC (dicyclohexylcarbodiimide);	CBZ (benzyloxycarbonyl);
Ac (acetyl);	atm (atmosphere);
TMSE (2-(trimethylsilyl)ethyl);	TMS (trimethylsilyl);
TIPS (triisopropylsilyl);	TBS ( <i>t</i> -butyldimethylsilyl);
DMAP (4-dimethylaminopyridine);	BSA (bovine serum albumin)
ATP (adenosine triphosphate);	HRP (horseradish peroxidase);
DMEM (Dulbecco's modified Eagle medium);	
HPLC (high pressure liquid chromatography);	
BOP (bis(2-oxo-3-oxazolidinyl)phosphinic chloride);	
TBAF (tetra- <i>n</i> -butylammonium fluoride);	
HBTU (O-Benzotriazole-1-yl- <i>N,N,N'</i> -tetramethyluroniumhexafluoro phosphate).	
HEPES (4-(2-hydroxyethyl)-1-piperazine ethane sulfonic acid);	
DPPA (diphenylphosphoryl azide);	
fHNO <sub>3</sub> (fuming HNO <sub>3</sub> ); and	
EDTA (ethylenediaminetetraacetic acid).	

All references to ether are to diethyl ether; brine refers to a saturated aqueous solution of NaCl. Unless otherwise indicated, all temperatures are expressed in °C

(degrees Centigrade). All reactions are conducted under an inert atmosphere at room temperature unless otherwise noted.

<sup>1</sup>H NMR spectra were recorded on a Varian VXR-300, a Varian Unity-300, a Varian Unity-400 instrument, a Bruker AVANCE-400, or a General Electric QE-300. Chemical shifts are expressed in parts per million (ppm,  $\delta$  units). Coupling constants are in units of Hertz (Hz). Splitting patterns describe apparent multiplicities and are designated as s (singlet), d (doublet), t (triplet), q (quartet), quint (quintet), m (multiplet), br (broad).

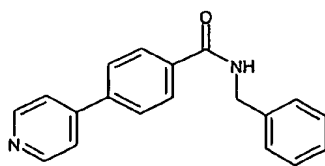
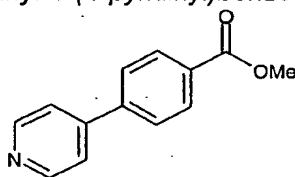
HPLC were recorded on a Gilson HPLC or Shimadzu HPLC system by the following conditions. Column: 50 X 4.6mm (id) stainless steel packed with 5 $\mu$ m Phenomenex Luna C-18 ; Flow rate: 2.0 mL/min; Mobile phase: A phase = 50mM ammonium acetate (pH 7.4), B phase = acetonitrile, 0-0.5min (A: 100%, B: 0%), 0.5-3.0 min (A:100-0%, B:0-100%), 3.0-3.5min (A: 0%, B: 100%), 3.5-3.7 min (A: 0-100%, B: 100-0%), 3.7-4.5 min (A: 100%, B: 0%); Detection : UV 254nm; Injection volume: 3 $\mu$ L.

Low-resolution mass spectra (MS) were recorded on a JOEL JMS-AX505HA, JOEL SX-102, or a SCIEX-APIiii spectrometer; LC-MS were recorded on a micromass 2MD and Waters 2690; high resolution MS were obtained using a JOEL SX-102A spectrometer. All mass spectra were taken under electrospray ionization (ESI), chemical ionization (CI), electron impact (EI) or by fast atom bombardment (FAB) methods. Infrared (IR) spectra were obtained on a Nicolet 510 FT-IR spectrometer using a 1-mm NaCl cell. Most of the reactions were monitored by thin-layer chromatography on 0.25 mm E. Merck silica gel plates (60F-254), visualized with UV light, 5% ethanolic phosphomolybdic acid or p-anisaldehyde solution. Flash column chromatography was performed on silica gel (230-400 mesh, Merck).



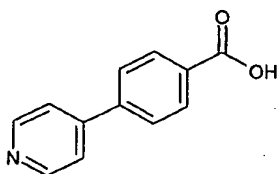
**Example 1**

Method A (see Scheme 1)

**N-benzyl-4-(4-pyridinyl)benzamide****(a) preparation of methyl 4-(4-pyridinyl)benzoate**

A mixture of methyl 4-bromobenzoate (0.200 g, 0.930 mmol), 4-pyridylboronic acid (0.172 g, 1.40 mmol), 2M aq.  $\text{Na}_2\text{CO}_3$  (0.70 mL, 1.40 mmol), dichlorobis(triphenylphosphine)palladium(II) (33 mg, 0.046 mmol) and DME (3.0 mL) was reacted at 175°C in a Creator™ microwave instrument for 10 min to give methyl 4-(4-pyridinyl)benzoate (0.20 g) as a solid.

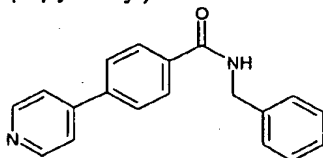
$^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  ppm 3.89 (s, 3H), 7.79 (d,  $J=6.2$  Hz, 2H), 7.97 (d,  $J=8.4$  Hz, 2H), 8.09 (d,  $J=8.4$  Hz, 2H), 8.69 (d,  $J=6.2$  Hz, 2H); MS  $m/z$  214 ( $M+1$ ) $^+$ .

**(b) preparation of 4-(4-pyridinyl)benzoic acid**

To a solution of methyl 4-(4-pyridinyl)benzoate (0.20 g, 0.93 mmol) and 1,4-dioxane (3 mL) was added 1M aq. lithium hydroxide (3 mL) and the mixture was stirred at rt for 2 days. The volatiles were removed by rotary evaporation under reduced pressure and water (1 mL) was added to the mixture. The mixture was acidified to pH 5 with concentrated HCl and then diluted with additional water (2 mL). The solids were collected by filtration and dried under vacuum to give 4-(4-pyridinyl)benzoic acid (0.17 g) as a solid.

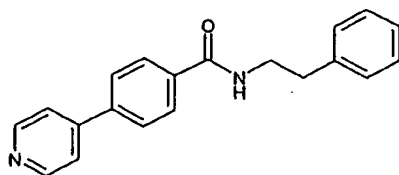
<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 7.78 (d,  $J$ =4.6 Hz, 2H), 7.94 (d,  $J$ =8.4 Hz, 2H), 8.07 (d,  $J$ =8.3 Hz, 2H), 8.67 (d,  $J$ =4.6 Hz, 2H), 13.13 (br s, 1H); MS  $m/z$  198 (M-1)<sup>-</sup>.

(c) preparation of *N*-benzyl-4-(4-pyridinyl)benzamide



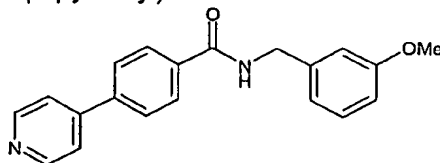
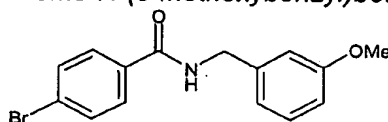
4-(4-Pyridinyl)benzoic acid (55 mg, 0.28 mmol) was added to a Bohdan filter cartridge containing Argonaut polystyrene-supported carbodiimide resin (519 mg, 0.69 mmol; 1.33 mmoles/g loading), HOBt (63 mg, 0.41 mmol), and DMF (3.0 mL). The mixture was shaken for 45 min and then benzylamine (26  $\mu$ L, 0.24 mmol) was added to the reaction and shaking continued for 2.5 days at rt. Next, Argonaut MP-carbonate resin (543 mg, 1.38 mmol) was added to scavenge the excess carboxylic acid and HOBt and shaking continued for 4 h. The filtrate was drained from the cartridge and DMF (2.0 mL) was added to the cartridge containing the resins. After shaking for 30 min, the filtrate was again drained and the combined filtrates were concentrated to dryness under high vacuum with heating. The residue was purified by using a pre-packed ISCO silica gel cartridge (4 gram) and eluted with a hexane : AcOEt linear solvent gradient (0% to 100% AcOEt) to give *N*-benzyl-4-(4-pyridinyl)benzamide (29 mg) as a solid.

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 4.54 (d,  $J$ =6.0 Hz, 2 H), 7.25-7.38 (m, 5 H), 7.82 (AB q,  $J$ =4.6 Hz, 2H), 7.96 (d,  $J$ =8.5 Hz, 2H), 8.08 (d,  $J$ =8.4 Hz, 2H), 8.70 (d,  $J$ =4.6 Hz, 2H); MS  $m/z$  289 (M+1)<sup>+</sup>.

**Example 2***N*-(2-phenylethyl)-4-(4-pyridinyl)benzamide

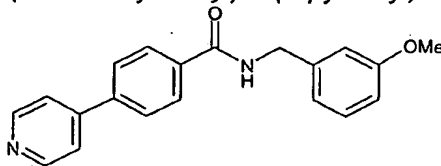
In a similar manner as Example 1c, 4-(4-pyridinyl)benzoic acid (55 mg, 0.28 mmol), Argonaut polystyrene-supported carbodiimide resin (519 mg, 0.69 mmol; 1.33 mmoles/g loading), HOBt (63 mg, 0.41 mmol), phenethylamine (30  $\mu$ L, 0.24 mmol) and DMF (3.0 mL) gave *N*-(2-phenylethyl)-4-(4-pyridinyl)benzamide (18 mg) as a solid.

$^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ )  $\delta$  ppm 2.90 (t,  $J=7.4$  Hz, 2H), 3.54 (m, 2H), 7.21-7.36 (m, 5 H), 7.80 (d,  $J=5.9$  Hz, 2H), 7.94 (d,  $J=8.6$  Hz, 2H), 8.00 (d,  $J=8.5$  Hz, 2H), 8.70 (d,  $J=6.0$  Hz, 2H), 8.72 (t, 1H); MS  $m/z$  303 ( $M+1$ ) $^+$ .

**Example 3****Method B (see Scheme 2)***N*-(3-methoxybenzyl)-4-(4-pyridinyl)benzamide**(a) preparation of 4-bromo-*N*-(3-methoxybenzyl)benzamide**

A mixture of 4-bromobenzoic acid (0.500 g, 2.49 mmol), HOBt (0.404 g, 2.99 mmol), EDC hydrochloride (0.573 g, 2.99 mmol) and DMF (10 mL) was stirred for 1 h at rt. Next, 3-methoxybenzylamine (0.351 mL, 2.74 mmol) was added and the reaction was stirred for 18 h at rt. The DMF was removed by rotary evaporation under reduced pressure and the oil was partitioned between AcOEt : water (50 mL : 10 mL). The phases were separated and the aqueous phase was extracted with AcOEt (25 mL). The combined organic layer was washed with 1N aq. sodium hydroxide (3 x 10 mL), water (2 x 10 mL), and then dried ( $\text{MgSO}_4$ ) for 20 hours. The volatiles were removed to give 4-bromo-*N*-(3-methoxybenzyl)benzamide (0.683 g) as an oil. MS  $m/z$  320/322 ( $M+1$ ) $^+$ .

(b) preparation of *N*-(3-methoxybenzyl)-4-(4-pyridinyl)benzamide

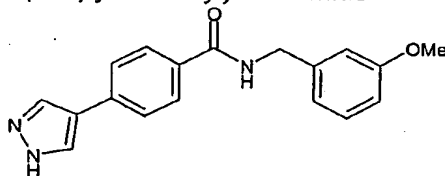


A mixture of 4-bromo-*N*-(3-methoxybenzyl)benzamide (150 mg, 0.468 mmol), 4-pyridylboronic acid (86 mg, 0.702 mmol), 2M aq. sodium carbonate (0.351 mL, 0.702 mmol), dichlorobis(triphenylphosphine)palladium(II) (16 mg, 0.023 mmol), DME (2.0 mL) and EtOH (1.0 mL) was placed into a 2 – 5 mL Emrys™ Process Vial from Personal Chemistry. The vial was capped and heated in a Personal Chemistry Creator™ microwave instrument at 175°C for 10 min. The reaction mixture was diluted with water (5.0 mL) and extracted with AcOEt (40 mL). The phases were separated and the organic layer was dried over MgSO<sub>4</sub> and the volatiles removed in the presence of 1.0 g of silica gel 60 (40-63  $\mu$ ). The pre-adsorbed material was chromatographed using a pre-packed ISCO silica gel cartridge (4 gram) and eluted with a hexane : AcOEt linear solvent gradient (0% to 100% AcOEt) to give *N*-(3-methoxybenzyl)-4-(4-pyridinyl)benzamide (91 mg) as a solid.

<sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  ppm 3.77 (s, 3H), 4.51 (d, *J*=5.9 Hz, 2H), 6.84-6.95 (m, 3H), 7.28 (t, *J*=8.1 Hz, 1H), 7.81 (d, *J*=4.6 Hz, 2H), 7.96 (d, *J*=8.4 Hz, 2H), 8.08 (d, *J*=8.5 Hz, 1H), 8.70 (d, *J*=4.6 Hz, 2H), 9.18 (t, *J*=6.0 Hz, 1H); MS *m/z* 319 (*M*+1)<sup>+</sup>.

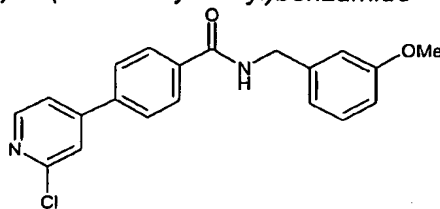
**Example 4**

*N*-(3-methoxybenzyl)-4-(1*H*-pyrazol-4-yl)benzamide



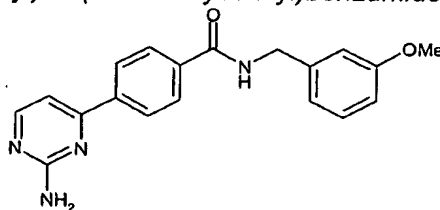
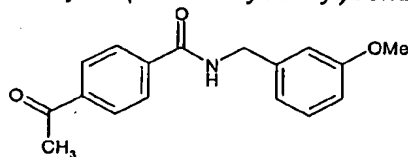
Prepared in a similar manner as described for Example 3b from 4-bromo-*N*-(3-methoxybenzyl)benzamide (150 mg, 0.468 mmol), 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*-pyrazole (0.182 g, 0.702 mmol), 2M aq. sodium carbonate (0.351 mL, 0.702 mmol), dichlorobis(triphenylphosphine)palladium(II) (16 mg, 0.023 mmol), DME (2.0 mL) and EtOH (1.0 mL) to give *N*-(3-methoxybenzyl)-4-(1*H*-pyrazol-4-yl)benzamide (56 mg) as a solid.

<sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  ppm 3.76 (s, 3H), 4.49 (d, *J*=6.0 Hz, 2H), 6.83-6.94 (m, 3H), 7.27 (t, *J*=8.1 Hz, 1H), 7.74 (d, *J*=8.4 Hz, 2H), 7.92 (d, *J*=8.4 Hz, 2H), 8.04 (br s, 1H), 8.34 (br s, 1H), 9.00 (t, *J*=6.0 Hz, 1H), 13.06 (br s, 1H); MS *m/z* 308 (*M*+1)<sup>+</sup>.

**Example 5****4-(2-chloro-4-pyridinyl)-N-(3-methoxybenzyl)benzamide**

Prepared in a similar manner as described for Example 3b from 4-bromo-N-(3-methoxybenzyl)benzamide (150 mg, 0.468 mmol), 2-chloro-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)pyridine (0.168 g, 0.702 mmol), 2M aq. sodium carbonate (0.351 mL, 0.702 mmol), dichlorobis(triphenylphosphine)palladium(II) (16 mg, 0.023 mmol), DME (2.0 mL) and EtOH (1.0 mL) to give 4-(2-chloro-4-pyridinyl)-N-(3-methoxybenzyl)benzamide (66 mg) as a solid.

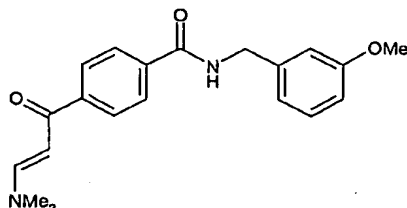
<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 3.77 (s, 3H), 4.51 (d,  $J$ =5.9 Hz, 2H), 6.84-6.95 (m, 3H), 7.28 (t,  $J$ =8.1 Hz, 1H), 7.86 (d,  $J$ =5.2 Hz, 1H), 7.97 (s, 1H), 8.02 (d,  $J$ =8.4 Hz, 2H), 8.08 (d,  $J$ =8.4 Hz, 2H), 8.54 (d,  $J$ =5.2 Hz, 1H), 9.21 (t,  $J$ =6.1 Hz, 1H); MS  $m/z$  353 (M+1)<sup>+</sup>.

**Example 6****Method C (see Scheme 3)****4-(2-amino-4-pyrimidinyl)-N-(3-methoxybenzyl)benzamide****(a) Preparation of 4-acetyl-N-(3-methoxybenzyl)benzamide**

Prepared in a similar manner as Example 3a using 4-acetylbenzoic acid (0.409 g, 2.49 mmol), HOBt (0.404 g, 2.99 mmol), EDC (0.573 g, 2.99 mmol), 3-methoxybenzylamine (0.351 mL, 2.74 mmol), and DMF (10 mL) to give 4-acetyl-N-(3-methoxybenzyl)benzamide (0.677 g) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 2.59 (s, 3H), 3.70 (s, 3H), 4.44 (d,  $J=5.9$  Hz, 2H), 6.78-6.88 (m, 3H), 7.22 (t,  $J=8.0$  Hz, 1H), 7.97 (d,  $J=8.4$  Hz, 2H), 8.01 (d,  $J=8.5$  Hz, 2H), 9.17 (t,  $J=5.8$  Hz, 1H); MS  $m/z$  284 (M+1)<sup>+</sup>.

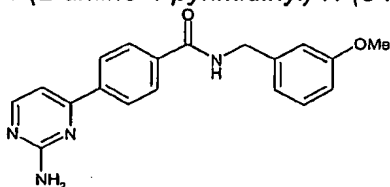
(b) *Preparation of 4-[(2E)-3-(dimethylamino)-2-propenoyl]-N-(3-methoxybenzyl)benzamide*



A mixture of 4-acetyl-N-(3-methoxybenzyl)benzamide (0.670 g, 2.36 mmol) and dimethylformamide dimethylacetal (3.1 mL, 24 mmol) was heated at reflux for 3 h and then cooled to rt. The volatiles were removed by rotary evaporation under reduced pressure and the residual solids were triturated in ether (50 mL), followed by filtration of the solids to give 4-[(2E)-3-(dimethylamino)-2-propenoyl]-N-(3-methoxybenzyl)benzamide (0.651 g) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 2.90/3.13 (2 x s, 6H), 3.70 (s, 3H), 4.43 (d,  $J=6.0$  Hz, 2H), 5.82 (d,  $J=12.3$  Hz, 1H), 6.77-6.87 (m, 3H), 7.21 (t,  $J=8.1$  Hz, 1H), 7.70 (d,  $J=12.3$  Hz, 1H), 7.90 (d,  $J=8.5$  Hz, 2H), 7.93 (d,  $J=8.5$  Hz, 2H), 9.08 (t,  $J=6.0$  Hz, 1H); MS  $m/z$  339 (M+1)<sup>+</sup>.

(c) *Preparation of 4-(2-amino-4-pyrimidinyl)-N-(3-methoxybenzyl)benzamide*



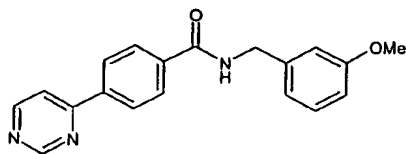
A small sphere of sodium (10 mg, 0.44 mmol) was dissolved in EtOH (4.0 mL). Guanidine hydrochloride (42 mg, 0.44 mmol) was added to the sodium ethoxide/EtOH and after 15 min 4-[(2E)-3-(dimethylamino)-2-propenoyl]-N-(3-methoxybenzyl)-benzamide (150 mg, 0.44 mmol) was added to the reaction and heated at reflux for 3 days. The cooled reaction was diluted with water (4 mL) and the precipitated solids were collected by filtration. The solids were rinsed with a small amount of water and then ether, then dried to give 4-(2-amino-4-pyrimidinyl)-N-(3-methoxybenzyl)-benzamide (116 mg) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 3.70 (s, 3H), 4.44 (d,  $J=5.8$  Hz, 2H), 6.70 (s, 2H), 6.78-6.88 (m, 3H), 7.17 (d,  $J=5.1$  Hz, 1H), 7.21 (t,  $J=8.0$  Hz, 1H), 7.97 (d,  $J=8.2$

Hz, 2H), 8.13 (d,  $J=8.3$  Hz, 2H), 8.31 (d,  $J=5.1$  Hz, 1H), 9.10 (t,  $J=5.9$  Hz, 1H); MS  $m/z$  335 ( $M+1$ )<sup>+</sup>.

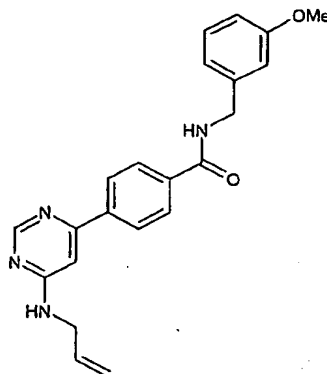
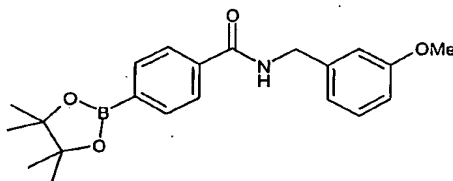
#### Example 7

##### *N*-(3-methoxybenzyl)-4-(4-pyrimidinyl)benzamide



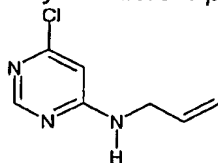
Prepared in a similar manner as Example 6c using formamidine hydro chloride (0.107 g, 1.33 mmol), K<sub>2</sub>CO<sub>3</sub> (0.307 g, 2.22 mmol), and 4-[(2E)-3-(dimethylamino)-2-propenoyl]-*N*-(3-methoxybenzyl)benzamide (150 mg, 0.44 mmol) in DMF (3.0 mL). The reaction was heated at 110°C for 7 days, then cooled and the mixture was filtered. The solids were rinsed with DMF (3 mL) and the combined filtrate was concentrated under reduced pressure. The resulting oil was partitioned between water : AcOEt (5 mL : 50 mL) and the phases were separated. The aqueous layer was extracted with AcOEt (50 mL) and the combined organic layer was dried over MgSO<sub>4</sub>. The organic layer was concentrated to dryness and then purified by column chromatography as in Example 3b to give *N*-(3-methoxybenzyl)-4-(4-pyrimidinyl)benzamide (65 mg) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  ppm 3.71 (s, 3 H), 4.45 (d,  $J=6.2$  Hz, 2H), 6.79 (d,  $J=6.9$  Hz, 2H), 6.80-6.89 (m, 2H), 7.22 (t,  $J=8.0$  Hz, 1H), 8.03 (d,  $J=8.2$  Hz, 2H), 8.16 (d,  $J=5.3$  Hz, 1H), 8.29 (d,  $J=8.2$  Hz, 2H), 8.88 (d,  $J=5.3$  Hz, 1H), 9.15 (t,  $J=5.9$  Hz, 1H), 9.26 (s, 1H); MS  $m/z$  320 ( $M+1$ )<sup>+</sup>.

**Example 8****Method D (see Scheme 4)****4-[6-(allylamino)-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide****(a) Preparation of N-(3-methoxybenzyl)-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzamide**

In a similar manner as described for Example 3a, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzoic acid (1.00 g, 4.03 mmol), HOBt (0.654 g, 4.84 mmol), EDC (0.928 g, 4.84 mmol), 3-methoxybenzylamine (0.567 mL, 4.43 mmol) and DMF (15 mL) gave N-(3-methoxybenzyl)-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzamide (1.25 g) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  ppm 1.27 (s, 12H), 3.69 (s, 3 H), 4.41 (d,  $J$ =5.8 Hz, 2H), 6.77-6.86 (m, 3H), 7.20 (t,  $J$ =8.0 Hz, 1H), 7.72 (d,  $J$ =8.0 Hz, 2H), 7.86 (d,  $J$ =8.1 Hz, 2H), 9.06 (t,  $J$ =6.1 Hz, 1 H).

**(b) Preparation of N-allyl-6-chloro-4-pyrimidinamine**

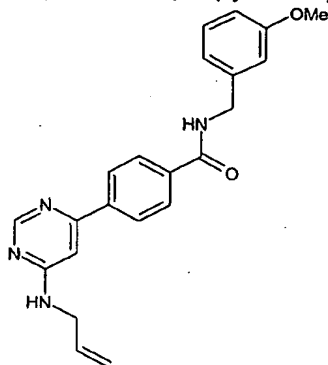
A mixture of 4,6-dichloropyrimidine (1.00 g, 6.71 mmol), allylamine (0.528 mL, 7.05 mmol), TEA (0.983 mL, 7.05 mmol) and THF (25 mL) was stirred at rt for 20 h. The volatiles were removed by rotary evaporation under reduced pressure and the residual oil was partitioned between water : AcOEt (10 mL : 50 mL). The phases were separated and the aqueous phase was extracted with AcOEt (50 mL). The



combined organic layer was dried over  $\text{MgSO}_4$  and concentrated to dryness to give N-allyl-6-chloro-4-pyrimidinamine (1.02 g) as a solid.

$^1\text{H}$  NMR (400 MHz,  $\text{DMSO-d}_6$ )  $\delta$  ppm 3.94 (br s, 2H), 5.10 (m, 2H), 5.84 (br s, 1H), 6.51 (br s, 1H), 7.86 (br s, 1H), 8.23 (br s, 1); MS  $m/z$  170 ( $\text{M}+1$ ) $^+$ .

(c) *preparation of 4-[6-(allylamino)-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide*

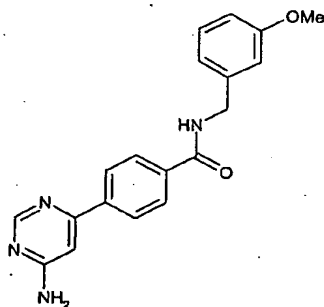


Prepared in a similar manner as described for Example 4 from N-allyl-6-chloro-4-pyrimidinamine (50 mg, 0.29 mmol), N-(3-methoxybenzyl)-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzamide (0.129 g, 0.35 mmol), 2M aq. sodium carbonate (0.175 mL, 0.35 mmol), dichlorobis(triphenylphosphine)palladium(II) (9.8 mg, 0.014 mmol), DME (1.0 mL) and EtOH (0.5 mL) to give 4-[6-(allylamino)-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide (24 mg) as a solid.

$^1\text{H}$  NMR (300 MHz,  $\text{DMSO-d}_6$ )  $\delta$  ppm 3.70 (s, 3H), 3.98 (br s, 2H), 4.44 (d,  $J=6.1$  Hz, 2H), 5.08 (d,  $J=5.3$  Hz, 1H), 5.18 (d,  $J=17.2$  Hz, 1H), 5.88 (m, 1H), 6.78 (d,  $J=7.1$  Hz, 1H), 6.80-6.88 (m, 2H), 6.99 (s, 1H), 7.21 (t,  $J=8.1$  Hz, 1H), 7.63 (t,  $J=5.7$  Hz, 1H), 7.97 (d,  $J=8.4$  Hz, 1H), 8.05 (br s, 2H), 8.49 (s, 1H), 9.09 (t,  $J=6.1$  Hz, 1H); MS  $m/z$  375 ( $\text{M}+1$ ) $^+$ .

**Example 9**

*4-[6-amino-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide*

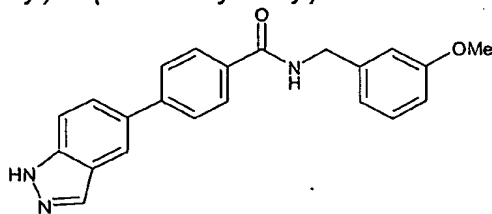


In a similar manner as Example 3b, a mixture of 4-[6-(allylamino)-4-pyrimidinyl]-N-(3-methoxybenzyl)benzamide (18 mg, 0.048 mmol), 1,3-dimethylbarbituric acid (7.5 mg, 0.048 mmol), tetrakis(triphenylphosphine)-palladium(0) (2.8 mg, 0.0024 mmol) and DCM (1.0 mL) was heated at 140°C for 25 min to give after chromatography 4-[6-amino-4-pyrimidinyl]-N-(3-methoxybenzyl)-benzamide (8.1 mg) as a solid.

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ ppm 3.71 (s, 3H), 4.44 (d, *J*=5.9 Hz, 2H), 6.79-6.89 (m, 3H), 6.93 (s, 1H), 6.95 (br s, 1H), 7.23 (t, *J*=8.0 Hz, 1H), 7.98 (d, *J*=8.4 Hz, 2H), 8.04 (d, *J*=8.2 Hz, 2H), 8.44 (s, 1H), 9.10 (t, *J*=6.0 Hz, 1H); MS *m/z* 335 (*M*+1)<sup>+</sup>.

### Example 10

4-(1*H*-indazol-5-yl)-N-(3-methoxybenzyl)benzamide



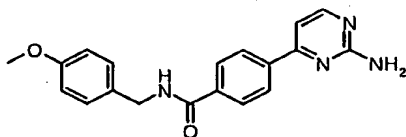
Prepared in a similar manner as described for Example 3b from 4-bromo-N-(3-methoxybenzyl)benzamide (150 mg, 0.468 mmol), 5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*-indazole (0.171 g, 0.702 mmol), 2M aq. sodium carbonate (0.351 mL, 0.702 mmol), dichlorobis(triphenylphosphine)palladium(II) (16 mg, 0.023 mmol), DME (2.0 mL) and EtOH (1.0 mL) to give 4-(1*H*-indazol-5-yl)-N-(3-methoxybenzyl)benzamide (4.8 mg) as a solid.

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ ppm 3.71 (s, 3H), 4.45 (d, *J*=5.9 Hz, 2H), 6.78-6.89 (m, 3H), 7.22 (t, *J*=8.2 Hz, 1H), 7.61 (d, *J*=8.8 Hz, 1H), 7.70 (d, *J*=8.8 Hz, 1H), 7.78 (d, *J*=8.5 Hz, 2H), 7.96 (d, *J*=8.4 Hz, 2H), 8.10 (d, *J*=10.4 Hz, 2H), 9.03 (t, *J*=6.0 Hz, 1H), 13.13 (br s, 1H); MS *m/z* 358 (*M*+1)<sup>+</sup>.

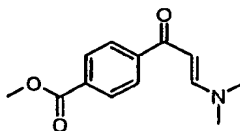
### Example 11

Method A (see Scheme 3a)

4-(2-Amino-4-pyrimidinyl)-N-[[4-(methyloxy)phenyl]methyl]benzamide

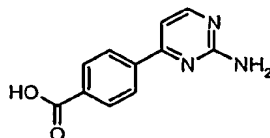


(a) Preparation of methyl 4-[(2*E*)-3-(dimethylamino)-2-propenoyl]benzoate



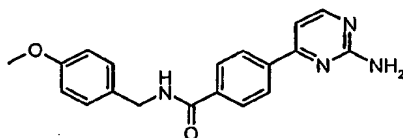
4-Acetylbenzoic acid (19 g, 116 mmol) was suspended in dimethylformamide dimethylacetal (43.8 mL) and the mixture was refluxed for 18 hours. A precipitate was formed and filtered. The solid was washed with ether to give 22.6 g of product. The filtrate was concentrated, extracted with EtOAc, and washed with water. The organic phase was washed with brine, dried ( $\text{MgSO}_4$ ), and concentrated to give 600 mg of product as a white solid. The two solids were combined to give 23.2 g (84%) of product as a white solid.

(b) *Preparation of 4-(2-amino-4-pyrimidinyl)benzoic acid*



To a solution of methyl 4-[(2E)-3-(dimethylamino)-2-propenoyl]benzoate (23.2 g, 97 mmol) in 2-methoxyethanol (400 mL) were added guanidine HCl salt (10.1 g, 107 mmol) and  $\text{K}_2\text{CO}_3$  (40 g, 291 mmol). The mixture was heated at reflux for 8 hours then room temperature over night. A precipitate was formed and filtered. This solid was dissolved in methanol and acetyl chloride (4.3 eq) was added to make the solution acidic. The solution was stirred for 3 hours, and the solvent was removed to give 46 g of solid (contained 21 g of KCl as maximum if the yield was quantitative). This solid was used for the next step without removing the salt.

(c) *Preparation of 4-(2-Amino-4-pyrimidinyl)-N-{[4-(methoxy)phenyl]methyl}benzamide*



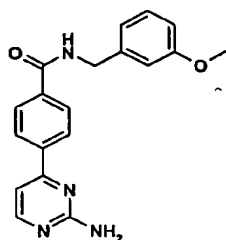
EDCI (98 mg, 0.51 mmol) was added to a solution of 4-(2-amino-4-pyrimidinyl)benzoic acid (100 mg, 0.47 mmol) in DMF (3 mL). After 10 minutes of stirring at room temperature, HOBT (100 mg, 0.74 mmol) was added. The mixture was stirred at room temperature for 10 minutes, then 1-[4-(methoxy)phenyl]methanamine (70 mg, 0.51 mmol) was added. A second portion of EDCI (98 mg) was added to the reaction mixture after 10 minutes of stirring. The reaction mixture was then stirred over night at room temperature. The reaction mixture was diluted with EtOAc, washed with water, brine, dried ( $\text{MgSO}_4$ ), and concentrated. The residue was dissolved in  $\text{CH}_2\text{Cl}_2$  and ether was added. A

precipitate was formed and filtered, and washed with ether to give 46 mg of product (30%). MS  $m/z$  335.2 ( $M+1$ )<sup>+</sup>.

All of the following compounds were made in a similar manner as described above in example 11.

**Example 12**

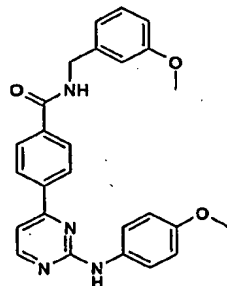
4-(2-amino-4-pyrimidinyl)-N-[[3-(methyloxy)phenyl]methyl]benzamide



MS  $m/z$  335.2 ( $M+1$ )<sup>+</sup>.

**Example 13**

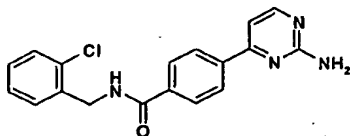
4-(2-[[4-(methyloxy)phenyl]amino]-4-pyrimidinyl)-N-[[3-(methyloxy)phenyl]methyl]benzamide



MS  $m/z$  441.1 ( $M+1$ )<sup>+</sup>.

**Example 14**

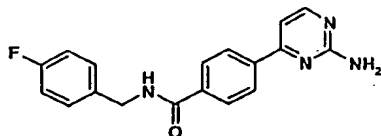
4-(2-amino-4-pyrimidinyl)-N-[(2-chlorophenyl)methyl]benzamide



MS  $m/z$  339.2 ( $M+1$ )<sup>+</sup>.

**Example 15**

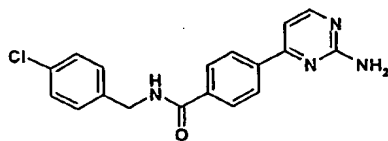
4-(2-amino-4-pyrimidinyl)-N-[(4-fluorophenyl)methyl]benzamide



MS  $m/z$  323.1 ( $M+1$ )<sup>+</sup>.

**Example 16**

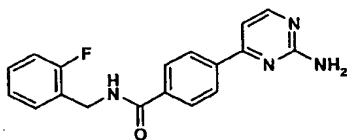
4-(2-amino-4-pyrimidinyl)-N-[(4-chlorophenyl)methyl]benzamide



MS m/z 339.1 (M+1)<sup>+</sup>.

**Example 17**

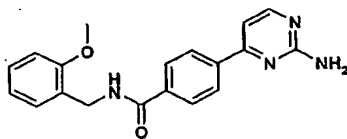
4-(2-amino-4-pyrimidinyl)-N-[(2-fluorophenyl)methyl]benzamide



MS m/z 323.1 (M+1)<sup>+</sup>.

**Example 18**

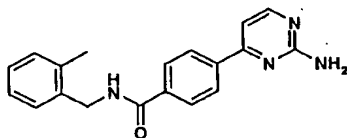
4-(2-amino-4-pyrimidinyl)-N-[[2-(methyloxy)phenyl]methyl]benzamide



MS m/z 335.1 (M+1)<sup>+</sup>.

**Example 19**

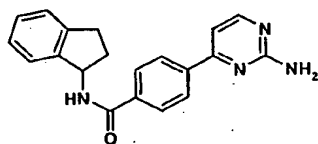
4-(2-amino-4-pyrimidinyl)-N-[(2-methylphenyl)methyl]benzamide



MS m/z 319.2 (M+1)<sup>+</sup>.

**Example 20**

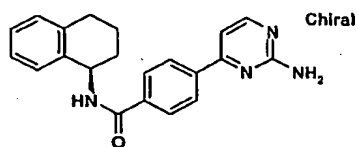
4-(2-amino-4-pyrimidinyl)-N-(2,3-dihydro-1H-inden-1-yl)benzamide



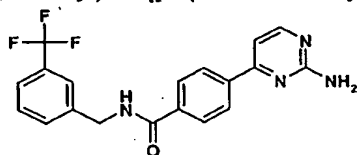
MS m/z 331.1 (M+1)<sup>+</sup>.

**Example 21**

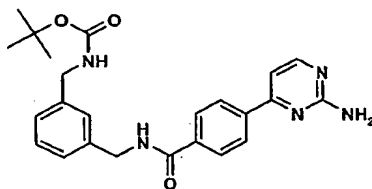
4-(2-amino-4-pyrimidinyl)-N-[(1R)-1,2,3,4-tetrahydro-1-naphthalenyl]benzamide

MS m/z 345.2 (M+1)<sup>+</sup>.**Example 22**

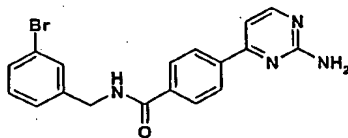
4-(2-amino-4-pyrimidinyl)-N-[(3-(trifluoromethyl)phenyl)methyl]benzamide

MS m/z 373.1 (M+1)<sup>+</sup>.**Example 23**

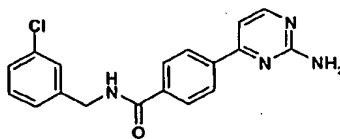
1,1-dimethylethyl({3-[(4-(2-amino-4-pyrimidinyl)phenyl)carbonyl]amino)methyl}phenyl)methyl carbamate

MS. m/z 434.1 (M+1)<sup>+</sup>.**Example 24**

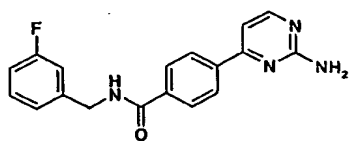
4-(2-amino-4-pyrimidinyl)-N-[(3-bromophenyl)methyl]benzamide

MS m/z 383.0 (M+1)<sup>+</sup>.**Example 25**

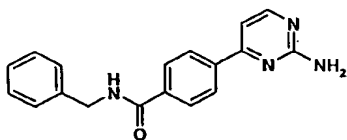
4-(2-amino-4-pyrimidinyl)-N-[(3-chlorophenyl)methyl]benzamide

**Example 26**

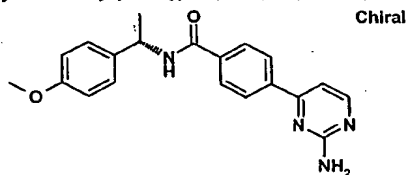
4-(2-amino-4-pyrimidinyl)-N-[(3-fluorophenyl)methyl]benzamide

MS m/z 323.1 (M+1)<sup>+</sup>.**Example 27**

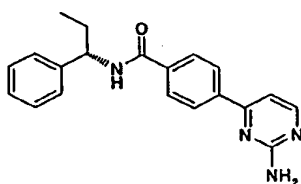
4-(2-amino-4-pyrimidinyl)-N-(phenylmethyl)benzamide

MS m/z 305.2 (M+1)<sup>+</sup>.**Example 28**

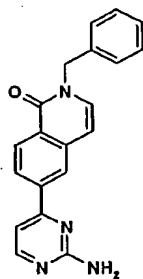
4-(2-amino-4-pyrimidinyl)-N-[(1S)-1-[4-(methyloxy)phenyl]ethyl]benzamide

MS m/z 349.2 (M+1)<sup>+</sup>.**Example 29**

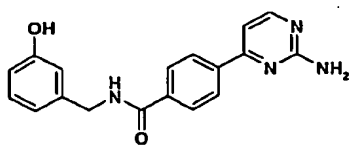
4-(2-amino-4-pyrimidinyl)-N-[(1S)-1-phenylpropyl]benzamide

MS m/z 333.2 (M+1)<sup>+</sup>.**Example 30**

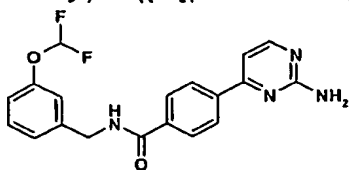
6-(2-amino-4-pyrimidinyl)-2-(phenylmethyl)-1(2H)-isoquinolinone

MS m/z 329.1 (M+1)<sup>+</sup>.**Example 31**

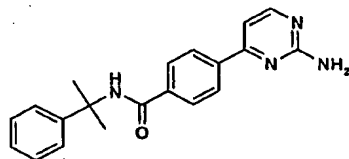
4-(2-amino-4-pyrimidinyl)-N-[(3-hydroxyphenyl)methyl]benzamide

MS m/z 321.1 (M+1)<sup>+</sup>.**Example 32**

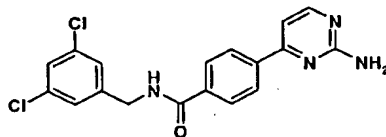
4-(2-amino-4-pyrimidinyl)-N-({3-[(difluoromethyl)oxy]phenyl}methyl)benzamide

MS m/z 371.1 (M+1)<sup>+</sup>.**Example 33**

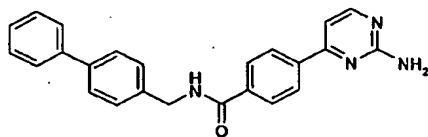
4-(2-amino-4-pyrimidinyl)-N-(1-methyl-1-phenylethyl)benzamide

MS m/z 333.2 (M+1)<sup>+</sup>.**Example 34**

4-(2-amino-4-pyrimidinyl)-N-[(3,5-dichlorophenyl)methyl]benzamide

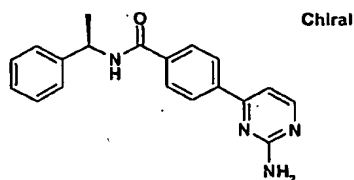
MS m/z 373.0 (M+1)<sup>+</sup>.**Example 35**

4-(2-amino-4-pyrimidinyl)-N-(4-biphenylmethyl)benzamide

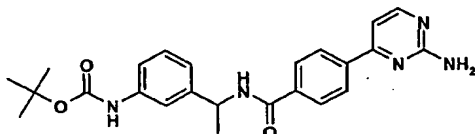
MS m/z 381.1 (M+1)<sup>+</sup>.**Example 36**

4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-phenylethyl]benzamide

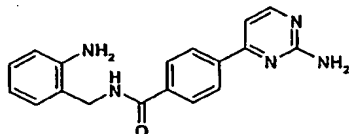


MS m/z 319.2 (M+1)<sup>+</sup>.**Example 37**

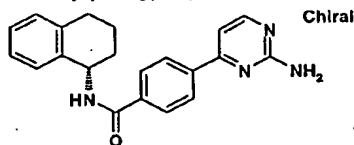
1,1-dimethylethyl{3-[1-({[4-(2-amino-4-pyrimidinyl)phenyl]carbonyl}amino)ethyl]phenyl}carbamate

MS m/z 434.1 (M+1)<sup>+</sup>.**Example 38**

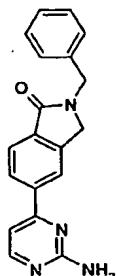
N-[(2-aminophenyl)methyl]-4-(2-amino-4-pyrimidinyl)benzamide

MS m/z 320.2 (M+1)<sup>+</sup>.**Example 39**

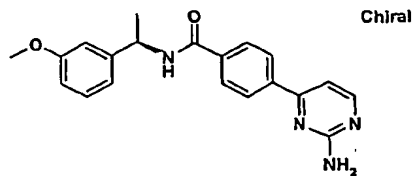
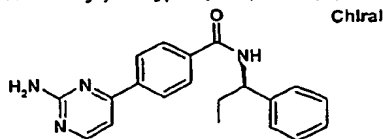
4-(2-amino-4-pyrimidinyl)-N-[(1S)-1,2,3,4-tetrahydro-1-naphthalenyl]benzamide

MS m/z 345.2 (M+1)<sup>+</sup>.**Example 40**

5-(2-amino-4-pyrimidinyl)-2-(phenylmethyl)-2,3-dihydro-1H-indol-1-one

MS m/z 317.1 (M+1)<sup>+</sup>.**Example 41**

4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-[3-(methyloxy)phenyl]ethyl]benzamide

MS m/z 349.1 (M+1)<sup>+</sup>.**Example 42***4-(2-amino-4-pyrimidinyl)-N-[(1R)-1-phenylpropyl]benzamide*MS m/z 333.2 (M+1)<sup>+</sup>.**ROCK kinase assay:**

ROCK inhibitor activity was determined using human recombinant ROCK1 kinase domain (amino acid 2-543) expressed in Sf9 cells (see WO9967283). The enzyme was purified using His-tag NTA column and Source15 HPLC chromatography. The assay of Rock-1 activity involved incubation with peptide substrate and ATP<sup>33</sup>, the subsequent incorporation of P<sup>33</sup> into the peptide was quantified by Scintillation Proximity Assay (SPA - Amersham Pharmacia).

For IC<sub>50</sub> determination, test compounds were typically dissolved at 10mM in 100% DMSO, with subsequent serial dilution in 100% DMSO. Compounds were typically assayed over an eleven point dilution range with a concentration in the assay of 50uM to 0.8nM, in 3-fold dilutions. IC<sub>50</sub> values were calculated by bespoke curve fitting software and then converted to pIC<sub>50</sub>.

Assays were performed in opaque, white walled, 384 well plates, in a total assay volume of 20ul. The assays contained: 1nM hROCK1; 1uM biotinylated peptide (biotin-Ahx-AKRRRLSSLRA-CONH<sub>2</sub>); 1uM ATP; 1.85kBq per well ATP(γ-<sup>33</sup>P); 25mM Hepes pH 7.4; 15mM MgCl<sub>2</sub>; 0.015% BSA. The reactions were incubated at 22°C for 120 minutes, then terminated by the addition of a 50ul solution containing 60mM EDTA and streptavidin PVT SPA beads. The SPA beads were added to a concentration of 0.14mg per well. The plates were allowed to incubate at 22°C for 10 minutes before centrifugation at 1500 rpm for 1 minute. P<sup>33</sup> incorporation was quantified by scintillation counting in a Packard TopCount.

All exemplified Examples were run with the recited assay and showed inhibitory activity versus Rock-1 with a pIC<sub>50</sub> of 5.0 or greater.